

# Real-Time Water Quality Deployment Report

# Voisey's Bay Network

June 10 to July 6, 2019



Government of Newfoundland & Labrador Department of Municipal Affairs and Environment Water Resources Management Division

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# **Real Time Water Quality Monitoring**

Staff with the Department of Municipal Affairs and Environment monitor the real-time web pages regularly.

This deployment report discusses water quality related events occurring at four stations in the Voisey's Bay Network: Reid Brook at Outlet to Reid Pond; Camp Pond Brook below Camp Pond; Tributary to Reid Brook; and Reid Brook below Tributary.

On June 10, 2019, Vale Environment staff deployed real-time water quality monitoring instruments at the four real-time stations in the Voisey's Bay network. Instruments were removed by Vale Environment Staff on July 6, 2019. This was the first deployment for the 2019 season.

# **Quality Assurance and Quality Control**

As part of the Quality Assurance and Quality Control protocol (QA/QC), an assessment of the reliability of data recorded by an instrument is made at the beginning and end of the deployment period. This procedure is based on the approach used by the United States Geological Survey.

At deployment and removal, a QA/QC instrument is temporarily deployed adjacent to the field instrument. Values for temperature, pH, conductivity, dissolved oxygen and turbidity are compared between the two instruments. Based on the degree of difference between parameters recorded by the field instrument and QA/QC instrument at deployment and at removal, a qualitative statement is made about the data quality (Table 1).

|                                 | Rank      |                |                |              |        |  |
|---------------------------------|-----------|----------------|----------------|--------------|--------|--|
| Parameter                       | Excellent | Good           | Fair           | Marginal     | Poor   |  |
| Temperature (oC)                | <=+/-0.2  | >+/-0.2 to 0.5 | >+/-0.5 to 0.8 | >+/-0.8 to 1 | <+/-1  |  |
| pH (unit)                       | <=+/-0.2  | >+/-0.2 to 0.5 | >+/-0.5 to 0.8 | >+/-0.8 to 1 | >+/-1  |  |
| Sp. Conductance (µS/cm)         | <=+/-3    | >+/-3 to 10    | >+/-10 to 15   | >+/-15 to 20 | >+/-20 |  |
| Sp. Conductance > 35 µS/cm (%)  | <=+/-3    | >+/-3 to 10    | >+/-10 to 15   | >+/-15 to 20 | >+/-20 |  |
| Dissolved Oxygen (mg/l) (% Sat) | <=+/-0.3  | >+/-0.3 to 0.5 | >+/-0.5 to 0.8 | >+/-0.8 to 1 | >+/-1  |  |
| Turbidity <40 NTU (NTU)         | <=+/-2    | >+/-2 to 5     | >+/-5 to 8     | >+/-8 to 10  | >+/-10 |  |
| Turbidity > 40 NTU (%)          | <=+/-5    | >+/-5 to 10    | >+/-10 to 15   | >+/-15 to 20 | >+/-20 |  |

Table 1: Ranking classifications for deployment and removal

It should be noted that the temperature sensor on any instrument is the most important. All other parameters can be broken down into three groups: temperature dependent; temperature compensated; and temperature independent. Because the temperature sensor is not isolated from the rest of the instrument, the entire instrument must be at the same temperature before the sensor will stabilize. The values may take some time to climb to the appropriate reading; if a reading is taken too soon it may not accurately portray the water body.

Deployment and removal comparison rankings for the Voisey's Bay Network stations are summarized in Table 2.

| Station<br>Voisey's Bay | Date    | Action     | Comparison Ranking |           |              |                     |           |
|-------------------------|---------|------------|--------------------|-----------|--------------|---------------------|-----------|
|                         |         |            | Temperature        | рН        | Conductivity | Dissolved<br>Oxygen | Turbidity |
| Reid Brook at Outlet    | June 10 | Deployment | Excellent          | Good      | Excellent    | Good                | Excellent |
|                         | July 6  | Removal    | Excellent          | Fair      | Excellent    | Fair                | Excellent |
| Camp Pond Brook         | June 10 | Deployment | Excellent          | Fair      | Excellent    | Excellent           | Excellent |
|                         | July 6  | Removal    | Excellent          | Good      | Good         | Excellent           | Poor      |
| Reid Brook below        | June 10 | Deployment | Excellent          | Excellent | Excellent    | Excellent           | Excellent |
| Tributary               | July 6  | Removal    | Excellent          | Good      | Excellent    | Good                | Excellent |
| Tributary to Reid Brook | June 10 | Deployment | Excellent          | Excellent | Excellent    | Excellent           | Excellent |
|                         | July 6  | Removal    | Excellent          | Marginal  | Excellent    | Good                | Good      |

Table 2: Comparison rankings for Voisey's Bay Network stations

#### Reid Brook at Outlet of Reid Pond

- At deployment, temperature, conductivity, and turbidity all ranked as 'excellent', while pH and dissolved oxygen were 'good'.
- At removal, temperature, conductivity, and turbidity all ranked as 'excellent', while pH and dissolved oxygen were 'fair'.

#### Camp Pond Brook below Camp Pond

- At deployment, temperature, conductivity, dissolved oxygen, and turbidity were all 'excellent', while pH was 'fair'.
- At removal, temperature and dissolved oxygen were 'excellent', while pH and conductivity were 'good'. Turbidity ranked as 'poor', which may be attributed to sediment build-up around the instrument.

#### **Reid Brook below Tributary**

- At deployment, all parameters ranked as 'excellent'.
- At removal, all parameters ranked as either 'excellent' or 'good'.

#### **Tributary to Reid Brook**

- At deployment, all parameters ranked as 'excellent'.
- At removal, temperature and conductivity were 'excellent', dissolved oxygen and turbidity were 'good', while pH was 'marginal'.

It is important to note that, in general, there are several conditions under which a less than ideal QA/QC ranking may be obtained. These include, but are not limited to: placement of the QA/QC sonde in relation to the field sonde; the amount of time each sonde is given to stabilize before readings are recorded; and deteriorating performance of one or more of the sensors.

# **Data Interpretation**

The following graphs and discussion illustrate significant water quality-related events from June 10<sup>th</sup> to July 6<sup>th</sup>, 2019 in the Voisey's Bay Real-Time Water Quality Monitoring Network.

With the exception of water quantity data (stage and flow), all data used in the preparation of the graphs and subsequent discussion below adhere to stringent QA/QC protocol. Water Survey of Canada is responsible for QA/QC of water quantity data. Corrected data can be obtained upon request.



Figure 1: Voisey's Bay Network Station Locations

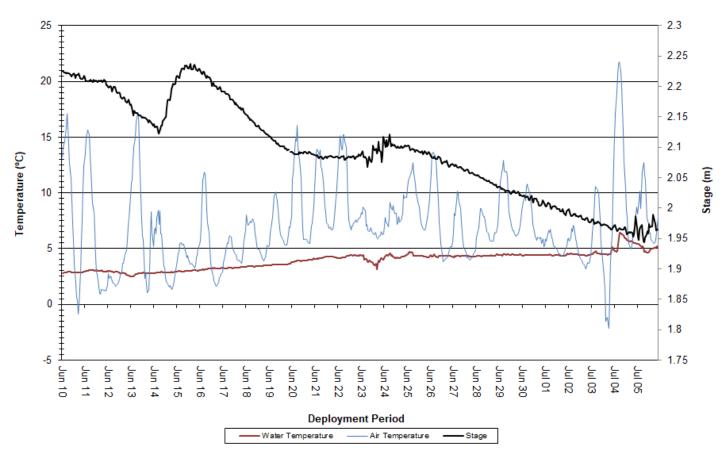
# **Reid Brook at Outlet of Reid Pond**

### Water Temperature

Over the deployment period, water temperature ranged from 2.49°C to 6.43°C, with a median value of 4.21°C (Figure 2). The very stable water temperature observed over the duration of deployment can be attributed to Reid Pond still being covered with ice. As evidenced in the graph below, air temperature fluctuates to a much greater extent each day compared to water temperature. Air temperature data was obtained from the Voisey's Bay airstrip weather station.

A noticeable increase in water temperature occurred on July 4<sup>th</sup> (Figure 2), which is likely a result of the warmer air temperatures occurring during the same time frame. This water body takes longer to acclimatize to changes in temperature as it has a much larger surface area compared to the brooks at the other RTWQ stations in this network.

Please note the stage data used below is raw data. It has not been corrected for backwater effect. WSC is responsible for QA/QC of water quantity data. Corrected data can be obtained upon request to WSC.



#### Reid Brook at Outlet of Reid Pond: Water and Air Temperature & Stage

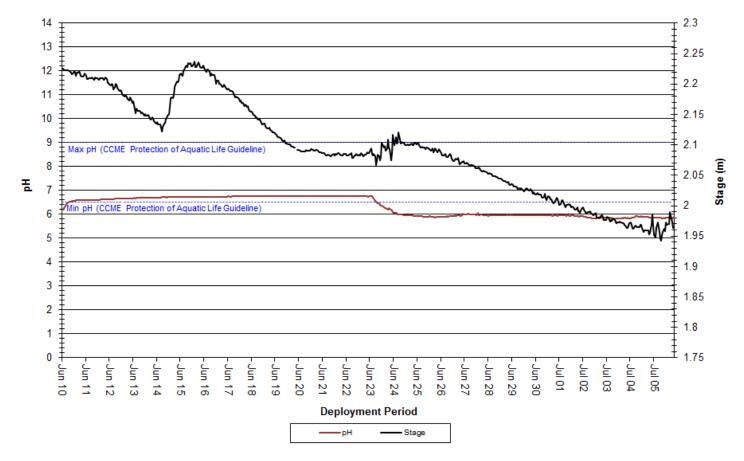
Figure 2: Water and Air Temperature & Stage at Reid Brook at Outlet of Reid Pond

# рΗ

Over the deployment period, pH values ranged from 5.81 pH units to 6.77 pH units, with a median value of 6.54 pH units (Figure 3).

pH levels were within the CCME's Guidelines for the Protection of Aquatic Life for the first half of deployment, after which they fell below the minimum guideline for the remainder of the deployment period. This decreased pH level corresponded closely with decreasing stage levels.

Please note the stage data used below is raw data. It has not been corrected for backwater effect. WSC is responsible for QA/QC of water quantity data. Corrected data can be obtained upon request to WSC.



## Reid Brook at Outlet of Reid Pond: pH & Stage

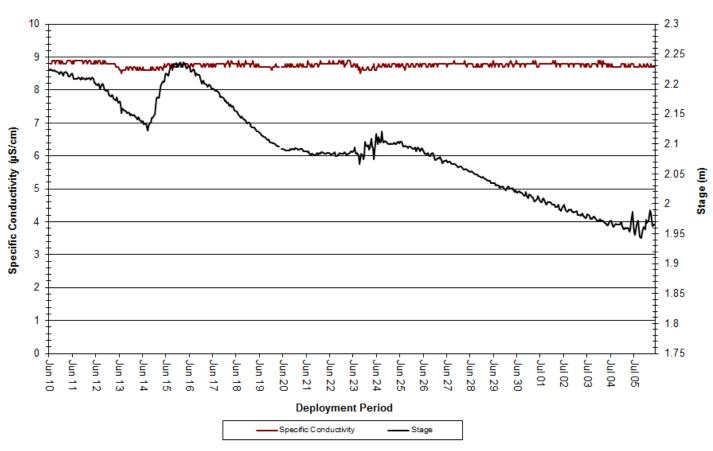
Figure 3: pH & Stage at Reid Brook at Outlet of Reid Pond

# **Specific Conductivity**

Over the deployment period, specific conductivity levels ranged from  $8.5\mu$ S/cm to  $8.9\mu$ S/cm, with a median value of  $8.8\mu$ S/cm. Conductivity at Reid Brook remained very stable. This is to be expected as this water body is pristine in nature and is far removed from any anthropogenic disturbances that could affect water quality.

The relationship between conductivity and stage level is generally inversed. When stage levels decrease, specific conductivity levels increase, as the decreased amount of water in the river system concentrates the solids that are present. Similarly, as stage levels rise, conductivity levels will dip in response. This relationship is not as evident at Reid Brook as it is at other stations in the Voisey's Bay network (Figure 4).

Please note the stage data used below is raw data. It has not been corrected for backwater effect. WSC is responsible for QA/QC of water quantity data. Corrected data can be obtained upon request to WSC.



#### Reid Brook at Outlet of Reid Pond: Specific Conductivity & Stage

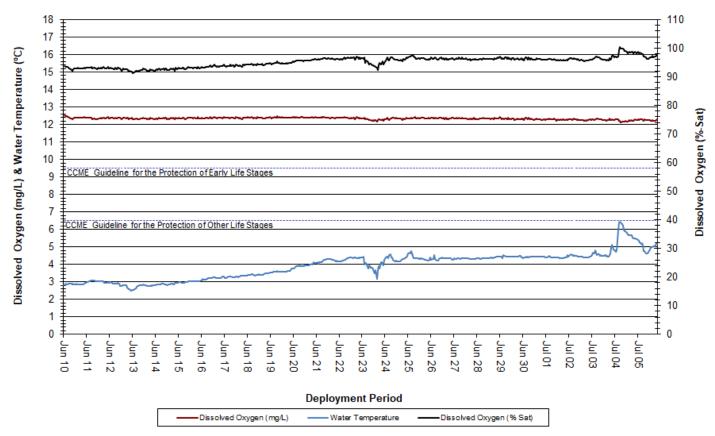
Figure 4: Specific Conductivity & Stage at Reid Brook at Outlet of Reid Pond

## **Dissolved Oxygen**

Over the deployment period, dissolved oxygen concentration levels ranged from 12.14mg/L to 12.52mg/L, with a median value of 12.35mg/L. Percent saturation levels for dissolved oxygen ranged from 91.4% saturation to 100.3% saturation, with a median value of 95.9% saturation (Figure 5).

The water quality instrument measures dissolved oxygen concentration (mg/L) with a dissolved oxygen probe. The instrument then calculates percent saturation (% Sat) taking into account water temperature.

Dissolved oxygen concentration values remained above the CCME's Guidelines for the Protection of Early Life Stages (9.5 mg/L) and Other Life Stages (6.5 mg/L) for the duration of deployment. Very stable water temperatures in Reid Pond over the course of deployment resulted in very little fluctuation in dissolved oxygen concentrations over the same period (Figure 5).



Reid Brook at Outlet of Reid Pond: Dissolved Oxygen Concentration and Saturation & Water Temperature

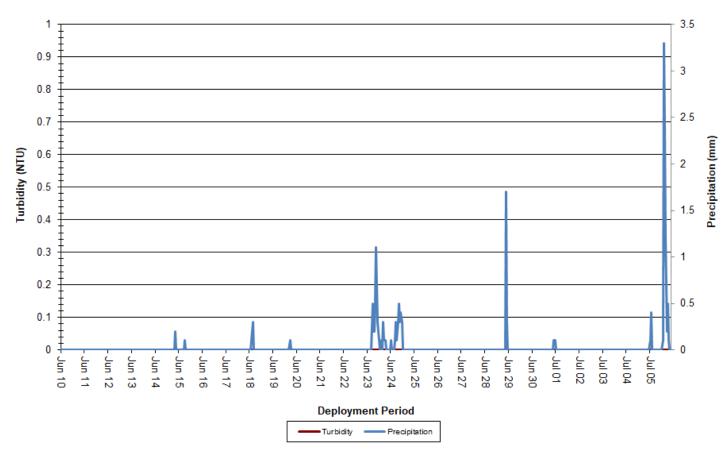
Figure 5: Dissolved Oxygen Concentration and Saturation & Water Temperature at Reid Brook at Outlet of Reid Pond

# Turbidity

Over the deployment period, turbidity levels remained stable at 0.0NTU (Figure 6). This indicates that there was very little background turbidity at this station during deployment.

All water bodies have a natural level of turbidity. A significant increase in turbidity is of concern when monitoring water quality. Higher turbidity readings would normally be expected during heavy rainfall or runoff events. Generally, turbidity levels increase for a short period of time and then return to within a baseline range. Turbidity values can also increase when there is a decrease in water level, which causes natural material in the water body to become concentrated.

Despite several precipitation events, turbidity levels remained at 0 NTU for the duration of deployment (Figure 6). This is not unusual for this station as it is pristine in nature and far removed from anthropogenic influences that may affect water quality.



## Reid Brook at Outlet of Reid Pond: Turbidity & Precipitation

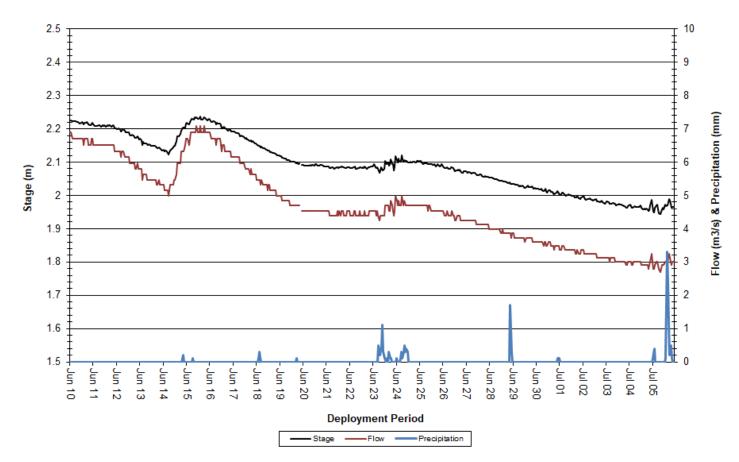
Figure 6: Turbidity & Precipitation at Reid Brook at Outlet of Reid Pond

# Stage, Flow & Precipitation

Stage is an important parameter, as it provides an estimate of water level at a station and can explain some of the events that are occurring with other parameters (e.g. specific conductivity, DO, and turbidity). Stage will generally increase during rainfall events (Figure 7) and during any surrounding snow or ice melt; however, direct snowfall will not cause a significant increase in stage.

Over the deployment period, stage values ranged from 1.94m to 2.24m, with a median value of 2.09m. Flow values ranged from 2.69m<sup>3</sup>/s to 7.08m<sup>3</sup>/s, with a median value of 4.54m<sup>3</sup>/s. Precipitation data was obtained from the Voisey's Bay airstrip weather station (Figure 7).

Please note the stage and flow data used below is raw data. It has not been corrected for backwater effect. WSC is responsible for QA/QC of water quantity data. Corrected data can be obtained upon request to WSC.



#### Reid Brook at Outlet of Reid Pond: Stage, Flow & Precipitation

Figure 7: Stage, Flow & Precipitation at Reid Brook at Outlet of Reid Pond

# Camp Pond Brook below Camp Pond

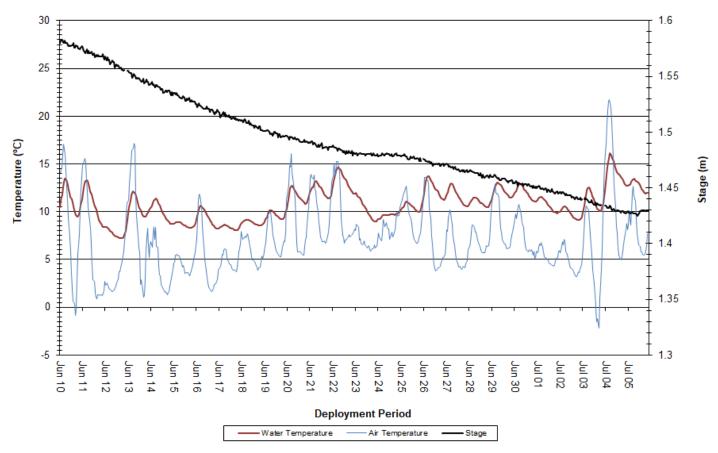
#### Water Temperature

Over the deployment period, water temperature ranged from 7.21°C to 16.09°C, with a median value of 10.75°C (Figure 8).

Water temperature at this station displays diurnal variations. There was a gradual increase in water temperature over the course of deployment. This is to be expected as air temperatures increased with the change from spring to summer (Figure 8). Air temperature data was obtained from the Voisey's Bay airstrip weather station.

Camp Pond Brook is sensitive to changes in ambient air temperature and fluctuates considerably depending on the weather and time of day. This station typically has the highest water temperatures and greatest fluctuations when compared to the other stations in the network.

Please note the stage data used below is raw data. It has not been corrected for backwater effect. WSC is responsible for QA/QC of water quantity data. Corrected data can be obtained upon request to WSC.



#### Camp Pond Brook below Camp Pond: Water and Air Temperature & Stage

Figure 8: Water and Air Temperature & Stage at Camp Pond Brook below Camp Pond

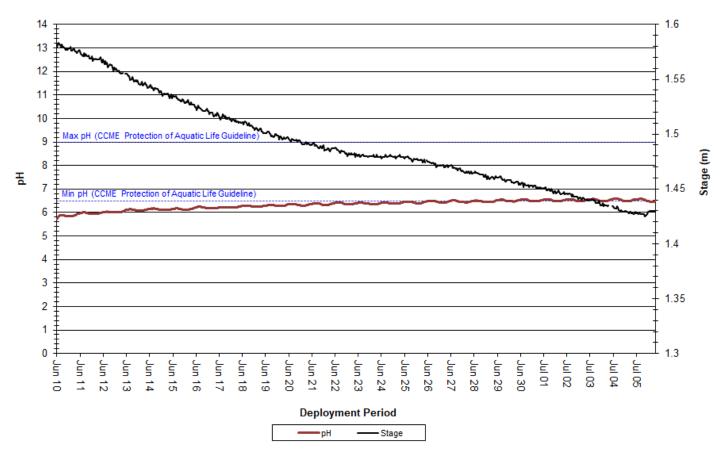
#### рΗ

Over the deployment period, pH values ranged from 5.70 pH units to 6.59 pH units, with a median value of 6.38 pH units (Figure 9).

pH levels increased slowly over the course of deployment, but remained below the CCME's Guidelines for the Protection of Aquatic Life for the majority of the deployment period.

Natural events such as rainfall and snow melt will alter the pH of a brook for a period of time - pH levels will decrease slightly during and after high stage levels. This is a natural process and may have contributed to lower pH levels observed at the beginning of the deployment period.

Please note the stage data used below is raw data. It has not been corrected for backwater effect. WSC is responsible for QA/QC of water quantity data. Corrected data can be obtained upon request to WSC.



#### Camp Pond Brook below Camp Pond: pH & Stage

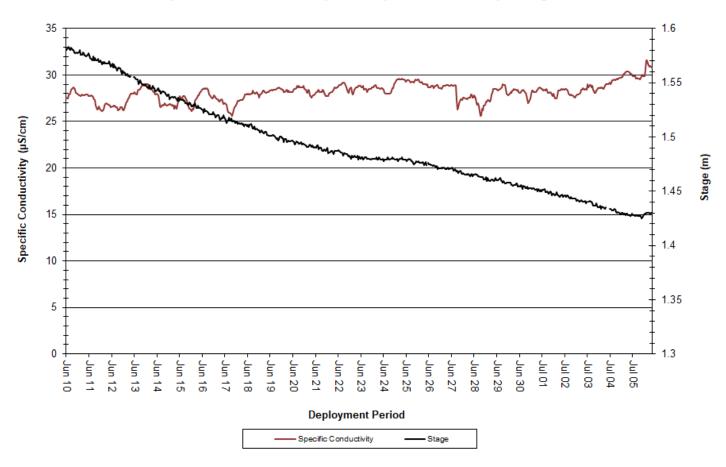
Figure 9: pH & Stage at Camp Pond Brook below Camp Pond

## **Specific Conductivity**

Over the deployment period, specific conductivity ranged from  $25.6\mu$ S/cm to  $31.6\mu$ S/cm, with a median value of  $28.3\mu$ S/cm (Figure 10).

Conductivity levels were variable but consistent across the deployment period as stage steadily decreased. A decrease in water level generally serves to concentrate suspended materials in the water column, in turn increasing specific conductivity; however, this association is not clearly visible in the graph below except towards the very end of deployment (Figure 10).

Please note the stage data used below is raw data. It has not been corrected for backwater effect. WSC is responsible for QA/QC of water quantity data. Corrected data can be obtained upon request to WSC.



#### Camp Pond Brook below Camp Pond: Specific Conductivity & Stage

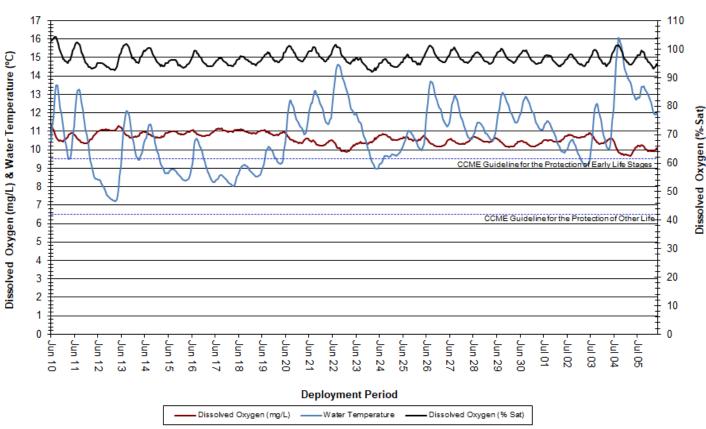
Figure 10: Specific Conductivity & Stage at Camp Pond Brook below Camp Pond

#### **Dissolved Oxygen**

Over the deployment period, dissolved oxygen concentration ranged from 9.68mg/L to 11.32mg/L, with a median value of 10.58mg/L. Saturation of dissolved oxygen ranged from 92.1% saturation to 104.4% saturation, with a median value of 96.3% (Figure 11).

Dissolved oxygen concentrations slowly decreased across the deployment period, while water temperature slowly increased. This observation is to be expected as water temperature directly influences the level of dissolved oxygen present in the water column; as water temperatures increase, dissolved oxygen concentrations decrease, and vice versa.

Dissolved oxygen concentrations remained above the CCME's Guideline for the Protection of Early Life Stages and Other Life Stages for the duration of the deployment period (Figure 11).



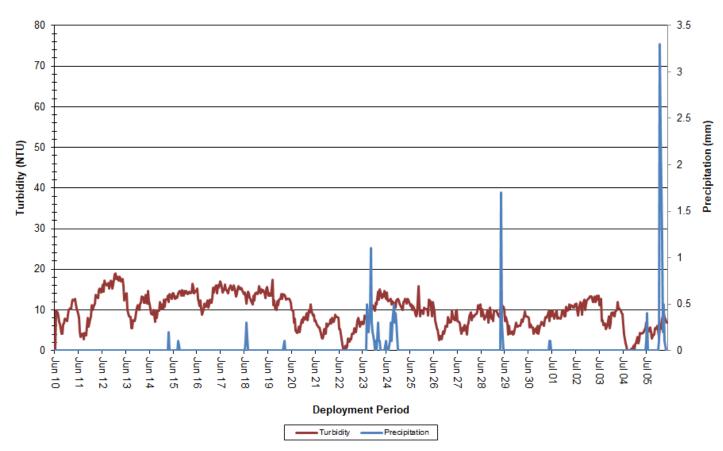
Camp Pond Brook below Camp Pond: Dissolved Oxygen (Concentration & Saturation) & Water Tempertaure

Figure 11: Dissolved Oxygen & Water Temperature at Camp Pond Brook below Camp Pond

# Turbidity

Over the deployment period, turbidity ranged from 0.0NTU to 18.9NTU, with a median value of 9.8NTU (Figure 12). A median value of 9.8NTU indicates that there was quite a bit of natural background turbidity at this station.

Higher turbidity levels are commonly observed at this station and are likely attributable to snow melt and subsequent runoff entering Camp Pond Brook (Figure 12). Precipitation data was obtained from the Voisey's Bay airstrip weather station.



## Camp Pond Brook below Camp Pond: Turbidity & Precipitation

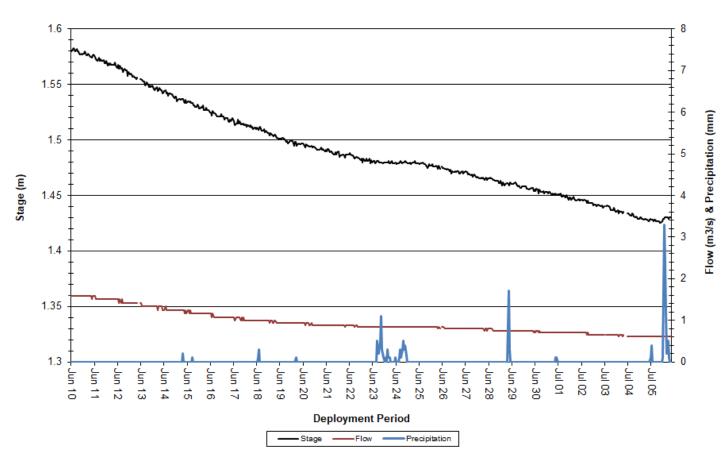
Figure 12: Turbidity & Precipitation at Camp Pond Brook below Camp Pond

## Stage, Flow and Precipitation

Over the deployment period, stage values ranged from 1.43m to 1.58m, with a median value of 1.48m. Stream flow values ranged from 0.61m<sup>3</sup>/s to 1.59m<sup>3</sup>/s, with a median value of 0.84m<sup>3</sup>/s. Precipitation data was obtained from the Voisey's Bay airstrip weather station (Figure 13).

Stage decreased steadily across the deployment period. A slight increase in stage was observed on July 6<sup>th</sup> and can be attributed to the observed rainfall event (Figure 13).

Please note the stage and flow data used below is raw data. It has not been corrected for backwater effect. WSC is responsible for QA/QC of water quantity data. Corrected data can be obtained upon request to WSC.



Camp Pond Brook below Camp Pond: Stage, Flow & Precipitation

Figure 13: Stage, Flow & Precipitation at Camp Pond Brook below Camp Pond

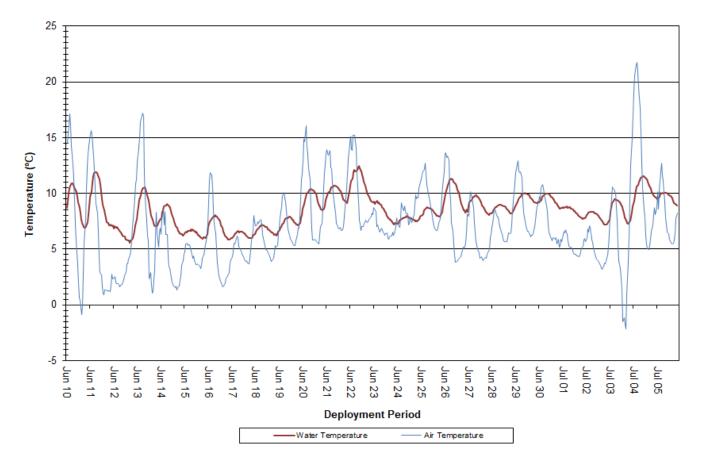
# **Reid Brook below Tributary**

### Water Temperature

Over the deployment period, water temperature ranged from 5.59°C to 12.43°C, with a median value of 8.49°C (Figure 14).

Water temperature at this station displays diurnal variations, and there was a gradual increase in the water temperature throughout the deployment period as spring changed to summer. This is to be expected as air temperatures also increased (Figure 14). Air temperature data was obtained from the Voisey's Bay airstrip weather station.

Stage data is not available because there was an issue with the transmission of data from this station during deployment. All other data was obtained via the RTWQ instrument's internal log file.



#### Reid Brook below Tributary: Water and Air Temperature

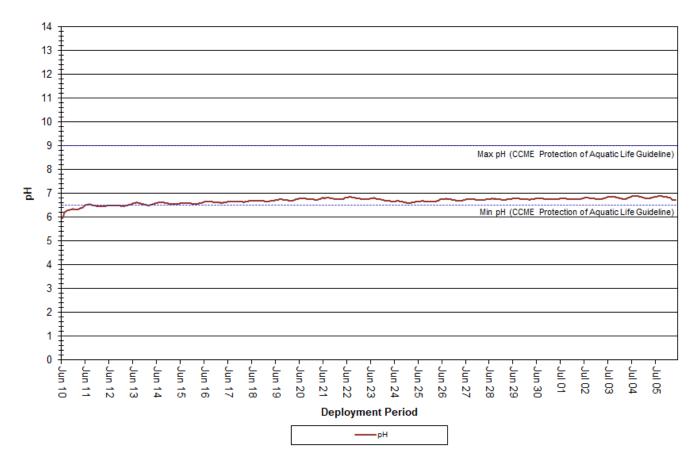
Figure 14: Water and Air Temperature at Reid Brook below Tributary

# рΗ

Over the deployment period, pH ranged from 5.97 pH units to 6.91 pH units, with a median value of 6.74 (Figure 15).

pH was below the CCME's Minimum Guideline for the Protection of Aquatic Life at the very beginning of the deployment period, but increased quickly and remained within the CCME's guidelines from June 13<sup>th</sup> onwards.

Stage data is not available because there was an issue with the transmission of data from this station during deployment. All other data was obtained via the RTWQ instrument's internal log file.



#### Reid Brook below Tributary: pH

Figure 15: pH at Reid Brook below Tributary

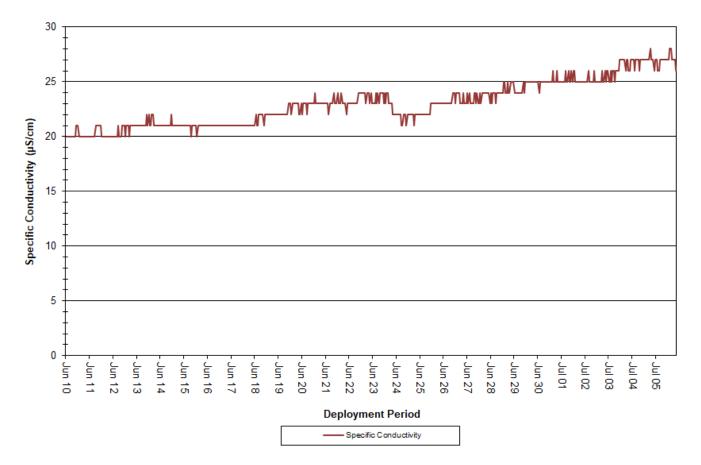
# **Specific Conductivity**

Over the deployment period, specific conductivity ranged from  $20.0\mu$ S/cm to  $28.0\mu$ S/cm, with a median value of  $23.0\mu$ S/cm (Figure 16).

Specific conductivity steadily increased over the course of deployment (Figure 16).

Specific conductivity and stage generally exhibit an inverse relationship: as one parameter increases, the other decreases. Increasing specific conductivity levels are likely due to decreasing water levels, which is characteristic of this station at this time of year; however, due to an equipment failure we do not have stage data to verify this relationship.

Stage data is not available because there was an issue with the transmission of data from this station during deployment. All other data was obtained via the RTWQ instrument's internal log file.



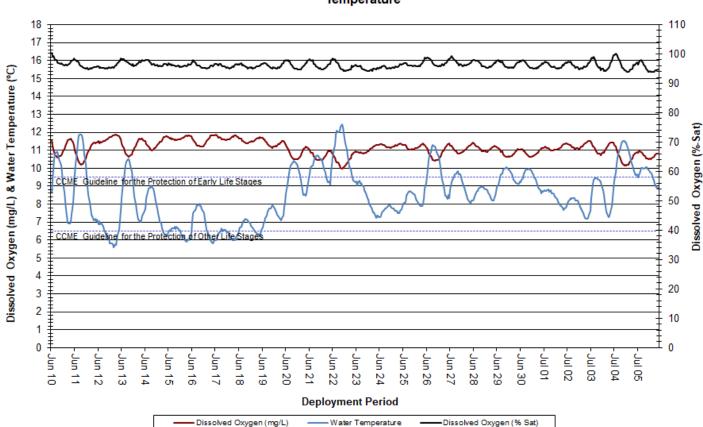
## Reid Brook below Tributary: Specific Conductivity

Figure 16: Specific Conductivity at Reid Brook below Tributary

# **Dissolved Oxygen**

Over the deployment period, dissolved oxygen concentration ranged from 9.98mg/L to 11.88mg/L, with a median value of 11.14mg/L. The saturation of dissolved oxygen ranged from 93.8% saturation to 100.5% saturation, with a median value of 96.1% (Figure 17).

Dissolved oxygen concentrations remained above the CCME's Guidelines for the Protection of Early Life Stages (9.5mg/L) and Other Life Stages (6.5 mg/L) for the duration of deployment. Dissolved oxygen concentration was fairly consistent over the course of deployment, with fluctuations closely connected to changes in water temperature. This is to be expected as water temperature and dissolved oxygen concentration generally exhibit an inverse relationship.



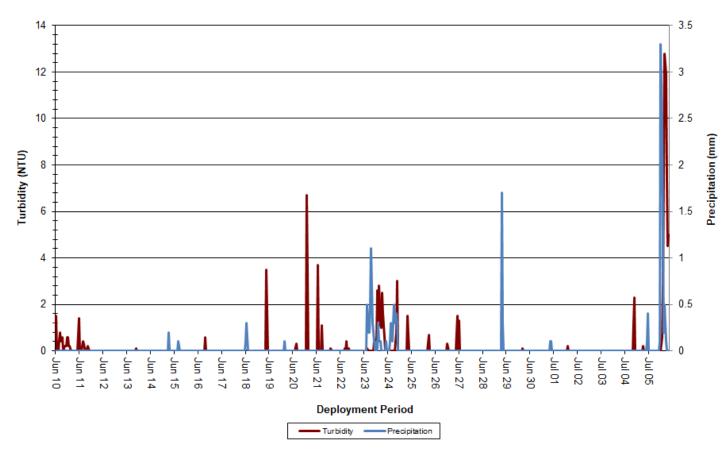
Reid Brook below Tributary: Dissolved Oxygen Concentration & Saturation and Water Temperature

Figure 17: Dissolved Oxygen & Water Temperature at Reid Brook below Tributary

# Turbidity

Over the deployment period, turbidity ranged from 0.0 NTU to 12.8 NTU, with a median value of 0.0 NTU (Figure 18). A median turbidity value of 0.0 NTU indicates that there was very little background turbidity at this station.

The majority of the turbidity events observed at this station closely correlated with rainfall events (Figure 18), which can cause mixing of solids in the water column. Precipitation data was obtained from the Voisey's Bay airstrip weather station.



#### Reid Brook below Tributary: Turbidity & Precipitation

Figure 18: Turbidity & Precipitation at Reid Brook below Tributary

# Stage and Flow

Stage and flow data are not available because there was an issue with the transmission of data from this station during deployment. All other data was obtained via the RTWQ instrument's internal log file.

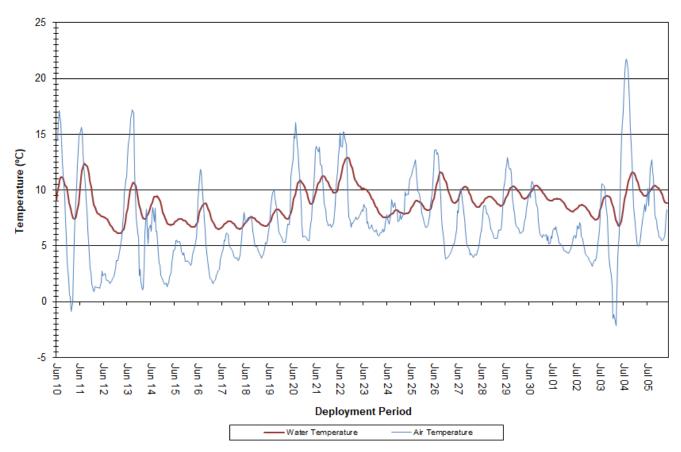
# **Tributary to Reid Brook**

#### Water Temperature

Over the deployment period, water temperature ranged from 6.12°C to 12.89°C, with a median value of 8.82°C (Figure 19). Streams and brooks are sensitive to changes in the ambient air temperature, thus water temperature will fluctuate considerably depending on the weather and the time of day. Air temperature fluctuates to a greater extent compared to water temperature. Air temperature data was obtained from the Voisey's Bay airstrip weather station.

Water temperature data displays a natural diurnal pattern. As expected, water temperatures increased gradually over the course of deployment with the change from spring to summer, and correlated closely with ambient air temperatures.

Stage data is not available because there was an issue with the transmission of data from this station during deployment. All other data was obtained via the RTWQ instrument's internal log file.



#### Tributary to Reid Brook: Water and Air Temperature

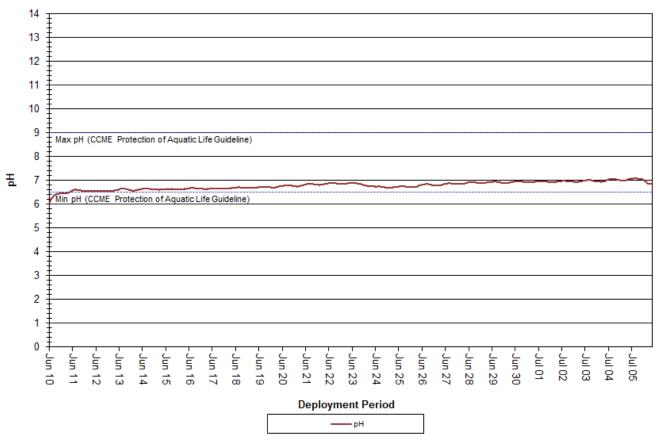
Figure 19: Water and Air Temperature at Tributary to Reid Brook

## рΗ

Over the deployment period, pH ranged from 6.11 pH units to 7.11 pH units, with a median value of 6.79 (Figure 20).

pH values reached the CCME's Minimum Guideline for the Protection of Aquatic Life shortly after the start of deployment, and remained within the acceptable range for the remainder of the deployment period.

Stage data is not available because there was an issue with the transmission of data from this station during deployment. All other data was obtained via the RTWQ instrument's internal log file.



## Tributary to Reid Brook: pH

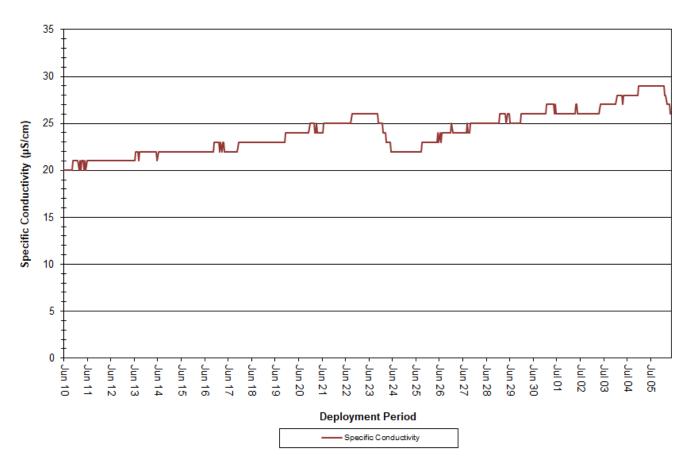
Figure 20: pH at Tributary to Reid Brook

# **Specific Conductivity**

Over the deployment period, specific conductivity ranged from 20.0μS/cm to 29.0μS/cm, with a median value of 24.0μS/cm (Figure 21).

Specific conductivity and stage generally exhibit an inverse relationship: as one parameter increases, the other decreases. Increasing specific conductivity levels are likely due to decreasing water levels, which is characteristic of this station at this time of year; however, due to an equipment failure we do not have stage data to verify this relationship.

Stage data is not available because there was an issue with the transmission of data from this station during deployment. All other data was obtained via the RTWQ instrument's internal log file.



#### Tributary to Reid Brook: Specific Conductivity

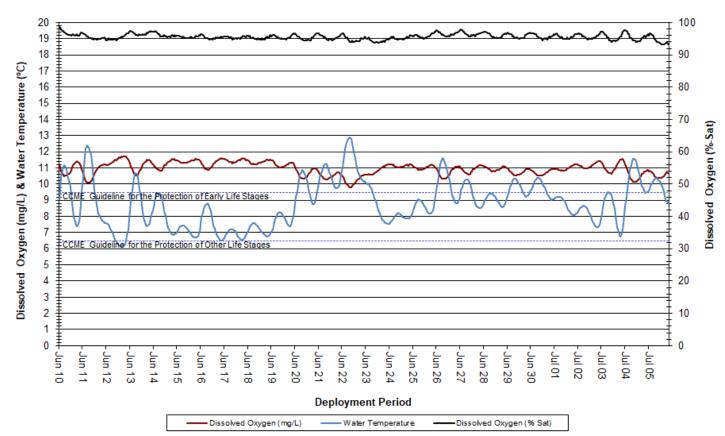
Figure 21: Specific Conductivity at Tributary to Reid Brook

## **Dissolved Oxygen**

Over the deployment period, dissolved oxygen concentration ranged from 9.80mg/L to 11.74mg/L, with a median value of 10.99mg/L. The saturation of dissolved oxygen ranged from 93.3% saturation to 98.6% saturation, with a median value of 95.6% (Figure 22).

Dissolved oxygen levels remained above the CCME's Guidelines for the Protection of Early and Other Life Stages for the duration of deployment.

Dissolved oxygen concentration displays a diurnal pattern. During nightfall, dissolved oxygen levels are higher as cooler temperatures allow for more DO to be stored in the water column. During the day, dissolved oxygen levels are lower. This is a result of warmer water temperatures and photosynthesis by aquatic plants, which decrease dissolved oxygen levels in the water column.



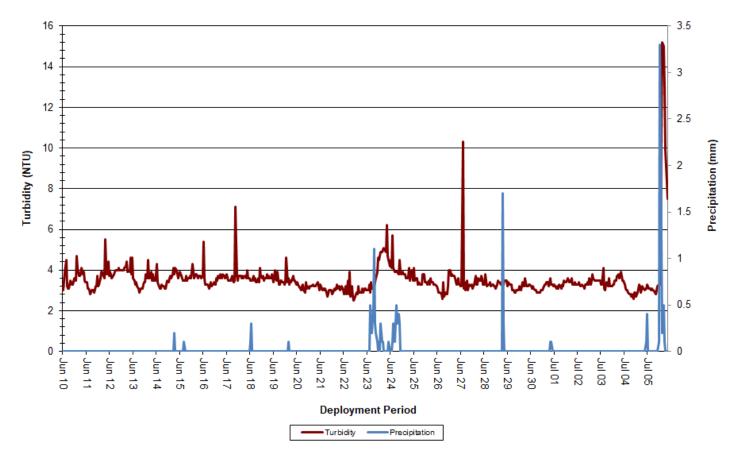
#### Tributary to Reid Brook: Dissolved Oxygen Concentration & Saturation and Water Temperature

Figure 22: Dissolved Oxygen & Water Temperature at Tributary to Reid Brook

# Turbidity

Over the deployment period, turbidity ranged from 2.5 NTU to 15.2 NTU, with a median value of 3.4 NTU (Figure 23). A median value of 3.4 NTU indicates that there was a small amount of natural background turbidity at this station.

There were a number of turbidity events at this station, with many of the larger turbidity events correlating with rainfall events (Figure 23). Rainfall generally causes an increase in water volume, which serves to stir up solid materials in the water column, in turn increasing turbidity. This site is particularly prone to variable turbidity, as it has a sandy-clay bottom that is easily disturbed by precipitation events.



#### Tributary to Reid Brook: Turbidity & Precipitation

Figure 23: Turbidity & Precipitation at Tributary to Reid Brook

# Stage and Flow

Stage and flow data are not available because there was an issue with the transmission of data from this station during deployment. All other data was obtained via the RTWQ instrument's internal log file.

# Conclusions

Water temperatures across all stations ranged from a minimum of 2.49°C at Reid Brook at Outlet of Reid Pond to a maximum of 16.09°C at Camp Pond Brook below Camp Pond. Overall, water temperature was increasing across the network. Stations at Camp Pond Brook, Tributary to Reid Brook, and Reid Brook below Tributary are more sensitive to changes in ambient air temperatures as these sites are brooks with continuously moving water. In contrast, Reid Brook at Outlet of Reid Pond is a large pond with a high surface area and deeper, slower-moving water, and was ice-covered for the majority of this deployment period.

pH values across all stations ranged from a minimum of 5.70pH units at Camp Pond Brook below Camp Pond to a maximum of 7.11pH units at Tributary to Reid Brook. pH values at all stations were relatively consistent across the deployment period.

Specific conductivity across all stations ranged from a minimum of  $8.5\mu$ S/cm at Reid Brook at Outlet of Reid Pond to a maximum of  $31.6\mu$ S/cm at Camp Pond Brook below Camp Pond. Conductivity values at Reid Brook at Outlet of Reid Pond were the lowest across the network. Camp Pond Brook below Camp Pond had the highest median value at  $28.3\mu$ S/cm, which is to be expected given the station's proximity to the Voisey's Bay mine site and increased potential for roadway runoff and other anthropogenic influences.

Dissolved oxygen levels across all stations ranged from a minimum of 9.68mg/L at Camp Pond Brook below Camp Pond to a maximum of 12.52mg/L at Reid Brook at Outlet of Reid Pond. Dissolved oxygen is generally higher at this time of year and varies diurnally as water temperature is greatly affected by ambient air temperature. Dissolved oxygen levels remained above the CCME's Guidelines for the Protection of Early and Other Life Stages at all stations for the duration of deployment.

Turbidity levels across all stations ranged from a minimum of 0.0 NTU at three stations to a maximum of 18.9NTU at Camp Pond Brook below Camp Pond. Turbidity levels showed natural increases and decreases generally corresponding to precipitation events.

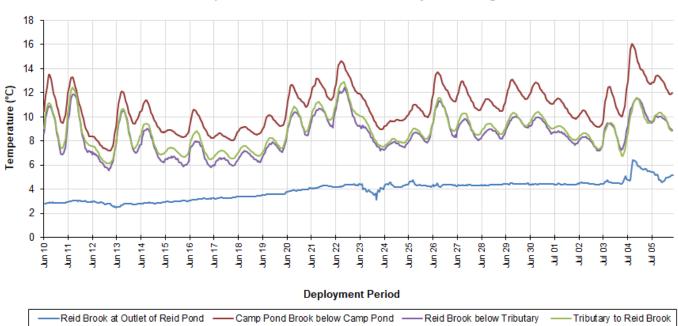
Air temperature and precipitation data were obtained from the Voisey's Bay weather station, which is located at the airstrip. This data appears to be quite accurate and no modifications were made.

Overall, the changes in water quality parameters over the course of this deployment can be explained by natural events. Camp Pond Brook below Camp Pond does have the potential for anthropogenic influences as the site is the closest to the inhabited area. It is important to note that during a change (a decrease or increase) in water quality, change only occurs for a short period of time and then water quality parameters return to baseline.

# References

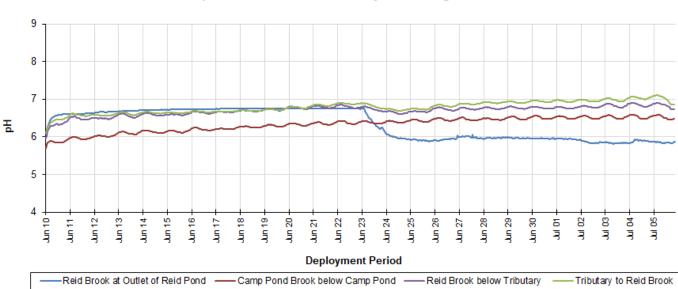
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**APPENDIX A: Comparison Graphs** 



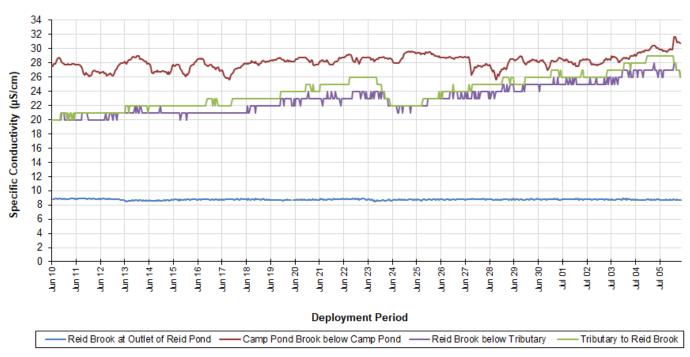
Water Temperature at Real-Time Water Quality Monitoring Stations

Figure A1: Comparison of Water Temperature (°C) between all Real-Time Stations in Voisey's Bay.



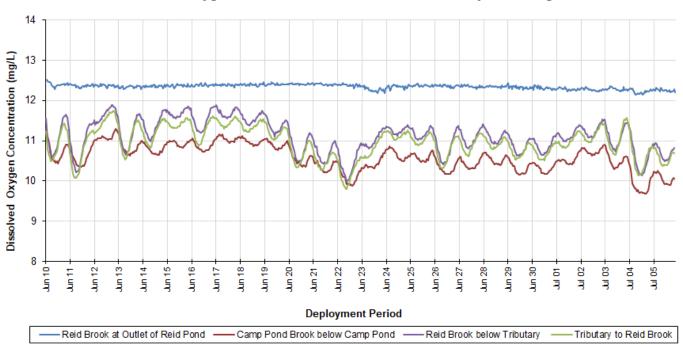
pH at Real-Time Water Quality Monitoring Stations

Figure A2: Comparison of pH between all Real-Time Stations in Voisey's Bay.



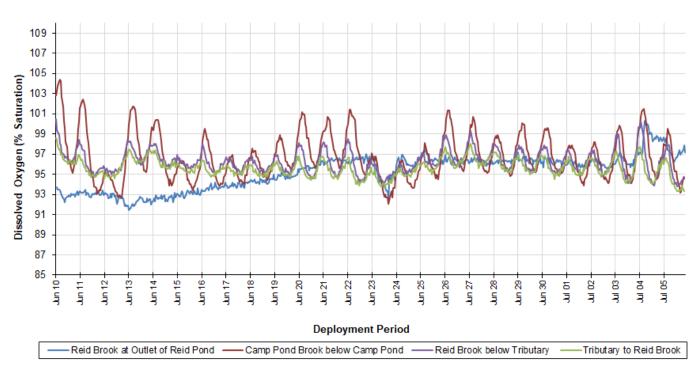
Specific Conductivity at Real-Time Water Quality Monitoring Stations

Figure A3: Comparison of Specific Conductivity (µS/cm) between all Real-Time Stations in Voisey's Bay.



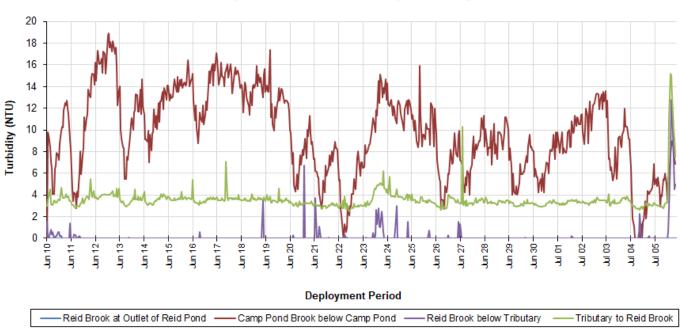
Dissolved Oxygen Concentration at Real-Time Water Quality Monitoring Stations

Figure A4: Comparison of Dissolved Oxygen (mg/L) between all Real-Time Stations in Voisey's Bay.



Dissolved Oxygen (% Saturation) at Real-Time Water Quality Monitoring Stations

Figure A5: Comparison of Dissolved Oxygen (% Sat) between all Real-Time Stations in Voisey's Bay.



Turbidity at Real-Time Water Quality Monitoring Stations

Figure A6: Comparison of Turbidity (NTU) between all Real-Time Stations in Voisey's Bay.

**APPENDIX B: Water Parameter Description** 

**Dissolved Oxygen**: The amount of Dissolved Oxygen (DO) (mg/L or % saturation) in the water is vital to the survival of aquatic organisms. The concentration of DO is affected by such things as water temperature, water depth and flow (e.g., aeration by rapids, riffles etc.), consumption by aerobic organisms, consumption by inorganic chemical reactions, consumption by plants during darkness, and production by plants during the daylight (CCME 2014).

**Flow:** Flow (m3/s) is a measure of how quickly a volume of water is displaced in streams, rivers, and other channels.

**pH:** pH is the measure of hydrogen ion activity and affects: (i) the availability of nutrients to aquatic life; (ii) the concentration of biochemical substances dissolved in water; (iii) the efficiency of hemoglobin in the blood of vertebrates; and (iv) the toxicity of pollutants. Changes in pH can be attributed to industrial effluence, saline inflows or aquatic organisms involved in the photosynthetic cycling of CO<sub>2</sub> (CCME 2014).

**Specific conductivity:** Specific conductivity ( $\mu$ S/cm) is a measure of water's ability to conduct electricity, with values normalized to a water temperature of 25°C. Specific conductance indicates the concentration of dissolved solids (such as salts) in the water, which can affect the growth and reproduction of aquatic life. Specific conductivity is affected by rainfall events, the composition of inflowing tributaries and their associated geology, saline inflow (e.g., road salt), agricultural run-off and industrial inputs (Swanson and Baldwin 1965).

**Stage:** Stage (m) is the elevation of the water surface and is often used as a surrogate for the more difficult to measure flow.

**Temperature:** Essential to the measurement of most water quality parameters, temperature (°C) controls most processes and dynamics of limnology. Water temperature is influenced by such things as ambient air temperature, solar radiation, meteorological events, industrial effluence, wastewater, inflowing tributaries, as well as water body size and depth (OTT Hydromet 2017).

**Total Dissolved Solids:** Total Dissolved Solids (TDS) (g/l) is a measure of alkaline salts dissolved in water or in fine suspension and can affect the growth and reproduction of aquatic life. It is affected by rainfall events, the composition of inflowing tributaries and their associated geology, saline inflow (e.g., road salt), agricultural run-off and industrial inputs (CCME 2014; Swanson and Baldwin 1965).

**Turbidity:** Turbidity (NTU) is a measure of the translucence of water and indicates the amount of suspended material in the water. Turbidity is caused by any substance that makes water cloudy (e.g., soil erosion, micro-organisms, vegetation, chemicals, etc.) and can correspond to precipitation events, high stage, and floating debris near the sensor (Sadar, 2017).

**APPENDIX C: Grab Sample Results** 



Lab Report Number: 1909867

| Cient:  |        | Department of Environme | ent                                    |   | COC Number:  | 844849   | )  |  |
|---|--------|-------------------------|--|---|--|--|--|--|
| Attention:  |        | Ms. Leona Hyde          |  |   | Date Reported:   | 2019-0   | 6-28   |  |
| Client Pro  | ject:  |                         |  |   | Date Submitted:  | 2019-06-17   |  |  |
| Purchase  | Order: | 2180014303              |  |   | Sample Matrix:   | Water  |  |  |
| <u>_AB ID</u><br>1433469<br>Sample comm   |        | 00                      | Client Sample ID<br>2019-6400-00-SI-SP | <u>Sample Date</u><br>2019-06-10  | <u>ANALYTE</u><br>Alkalinity as CaCO3<br>Bromide<br>Chloride<br>Colour | UNIT<br>mg/L<br>mg/L<br>mg/L<br>TCU                                      | <u>MRL</u><br>5<br>0.25<br>1<br>2  | <u>RESUL1</u><br><5<br><0.25<br><1<br>10 |
| Redi Brook @ Outlet<br><u>Sample comment:</u><br>Holding time for N-NO2, N-NO3 analysis was exceeded.<br><u>Report comment:</u> |        |                         |  | Conductivity<br>Dissolved Organic Carbon<br>Fluoride<br>Hardness as CaCO3<br>N-NH3 (Ammonia)<br>N-NO2 (Nitrite)<br>N-NO3 (Nitrate)<br>pH<br>Sulphate<br>Total Dissolved Solids (COND - CALC)<br>Total Kjeldahl Nitrogen | uS/cm<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/   | 5<br>0.5<br>0.10<br>1<br>0.010<br>0.10<br>0.10<br>1.00<br>1<br>1<br>0.15 | 14<br>1.9<br><0.10<br>2<br><0.010<br><0.10<br><0.10<br>6.53<br>1<br>9<br><0.15 |  |
|   |        |                         |  |   | Total Organic Carbon<br>Turbidity<br>Aluminum                          | mg/L<br>NTU<br>mg/L  | 0.5<br>0.1<br>0.01   | 2.2<br>0.5<br>0.05                       |

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Methods references and/or additional QA/QC information available on request.

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Lab Report Number: 1909867

| Cient:  | Department of Environment |   |                                  |  | COC Number:     | 84484  | 9  |   |
|---|---------------------------|---|----------------------------------|--|-----------------|--|--|---|
| Attention:  | Ms. Leona Hyde            |   |                                  |  | Date Reported:  | 2019-06-28<br>2019-06-17   |  |   |
| <u>AB ID</u> <u>Supply / De</u><br>1433469 WS-S-000<br>Redi Broo<br>Sample comment: |                           |   |                                  |  | Date Submitted: |  |  |   |
| Purchase Order:   | 2180014303                |   |                                  |  | Sample Matrix:  | Water  |  |   |
| 1433469 WS-S-<br>Redi E<br>Sample comment:  |                           | <u>Client Sample ID</u><br>2019-6400-00-SI-SP | <u>Sample Date</u><br>2019-06-10 | ANALYTE<br>Antimony<br>Arsenic<br>Barium<br>Boron<br>Calcium<br>Calcium<br>Cadmium<br>Chromium<br>Copper<br>Iron<br>Lead |                 | UNIT<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L | MRL<br>0.0005<br>0.001<br>0.01<br>1<br>0.0001<br>0.001<br>0.001<br>0.03<br>0.001 | RESULT<br><0.0005<br><0.001<br><0.01<br><0.001<br>1<br><0.0001<br><0.001<br><0.03<br><0.001 |
|   |                           |   |                                  | Magnesium<br>Manganese<br>Mercury<br>Nickel<br>Potassium<br>Selenium<br>Sodium<br>Strontium                              |                 | mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L                 | 1<br>0.01<br>0.0001<br>0.005<br>1<br>0.001<br>2<br>0.001                         | <1<br><0.01<br><0.0001<br><0.005<br><1<br><0.001<br><2<br>0.005                             |

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Methods references and/or additional QA/QC information available on request.

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| 0.0  | eurofi | 113 |
|      |        |     |

| Cient:                   |         | Department of Environr                   | nent  |                                  | COC Nun                                  | nber:   | 844849               |                             |                           |
|--------------------------|---------|--|---|----------------------------------|--|---------|----------------------|-----------------------------|---------------------------|
| Attention:               |         | Ms. Leona Hyde                           |   |                                  | Date Rep                                 | orted:  | 2019-06-             | -28                         |                           |
| Client Pro               | ject:   |  |   |                                  | Date Sub                                 | mitted: | 2019-06-             | -17                         |                           |
| Purchase                 | Order:  | 2180014303                               |   |                                  | Sample M                                 | latrix: | Water                |                             |                           |
| <u>LAB ID</u><br>1433469 | WS-S-00 | <u>Description</u><br>000<br>ok @ Outlet | <u>Client Sample ID</u><br>2019-6400-00-SI-SP | <u>Sample Date</u><br>2019-06-10 | ANALYTE<br>Uranium<br>Zinc<br>Dhaanbarua | n       | JNIT<br>ng/L<br>ng/L | <u>MRL</u><br>0.001<br>0.01 | RESULT<br><0.001<br><0.01 |
| Sample comn              | nent:   |  |   |                                  | Phosphorus<br>Total Suspended Solids     |         | ng/L<br>ng/L         | 0.002<br>2                  | <0.002<br><2              |

Holding time for N-NO2, N-NO3 analysis was exceeded.

Report comment:

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Lab Report Number:

1909867

| Cient:  | Department of Environme | nt  |                           | COC Number:   | 844849  | )  |   |
|---|-------------------------|---|---------------------------|---|---|--|---|
| Attention:  | Ms. Leona Hyde          |   |                           | Date Reported:  | 2019-0  | 6-28   |   |
| Attention:<br>Client Project:<br>Purchase Order:<br>AB ID Supply / De<br>433470 WS-S-000<br>Camp Por<br>ample comment:<br>Holding time for N-NO2, N |                         |   |                           | Date Submitted:   | 2019-06-17  |  |   |
| Purchase Order:   | 2180014303              |   |                           | Sample Matrix:  | Water   |  |   |
| 1433470 WS-S-00<br>Camp Pc<br>Sample comment:   | 00                      | <u>Client Sample ID</u><br>2019-6401-00-SI-SP | Sample Date<br>2019-06-10 | ANALYTE<br>Alkalinity as CaCO3<br>Bromide<br>Chloride<br>Colour<br>Conductivity<br>Dissolved Organic Carbon<br>Fluoride<br>Hardness as CaCO3<br>N-NH3 (Ammonia)<br>N-NO2 (Nitrite)<br>N-NO3 (Nitrate)<br>pH<br>Sulphate<br>Total Dissolved Solids (COND - CALC)<br>Total Kjeldahl Nitrogen<br>Total Organic Carbon<br>Turbidity | UNIT<br>mg/L<br>mg/L<br>TCU<br>uS/cm<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/ | MRL<br>5<br>0.25<br>1<br>2<br>5<br>0.5<br>0.10<br>1<br>0.010<br>0.10<br>0.10<br>1.00<br>1<br>1<br>0.15<br>0.5<br>0.1 | RESULT<br>7<br><0.25<br>2<br>26<br>33<br>3.8<br><0.10<br>7<br><0.010<br><0.10<br><0.10<br><0.10<br>6.88<br>3<br>21<br><0.15<br>3.6<br>0.8<br>0.07 |

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Lab Report Number: 1909867

| Cient:  |  | Department of Environr  | nent  |                                  |  | COC Number:     | 84484  | 9   |   |
|---|--|---|---|----------------------------------|--|-----------------|--|---|---|
| Attention:  |  | Ms. Leona Hyde  |   |                                  |  | Date Reported:  | 2019-06-28<br>2019-06-17   |   |   |
| Client Proj   | ect:                                     |   |   |                                  |  | Date Submitted: |  |   |   |
| Purchase (  | Order:                                   | 2180014303  |   |                                  |  | Sample Matrix:  | Water  |   |   |
| LAB ID<br>1433470<br>Sample comme<br>Holding time<br>Report comme | WS-S-00<br>Camp Po<br>ent:<br>for N-NO2, | Description<br>000<br>ond Brook<br>N-NO3 analysis was exceeded. | <u>Client Sample ID</u><br>2019-6401-00-SI-SP | <u>Sample Date</u><br>2019-06-10 | ANALYTE<br>Antimony<br>Arsenic<br>Barium<br>Boron<br>Calcium<br>Cadmium<br>Chromium<br>Chromium<br>Copper<br>Iron<br>Lead<br>Magnesium<br>Manganese<br>Mercury<br>Nickel |                 | UNIT<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L | MRL<br>0.0005<br>0.001<br>0.01<br>1<br>0.001<br>0.001<br>0.001<br>0.03<br>0.001<br>1<br>0.03<br>0.001<br>1<br>0.01<br>0.0 | RESULT<br><0.0005<br><0.001<br><0.01<br>3<br><0.0001<br><0.001<br>0.003<br>0.19<br><0.001<br><1<br>0.01<br><0.0001<br>0.024 |
|   |  |   |   |                                  | Potassium<br>Selenium<br>Sodium<br>Strontium   |                 | mg/L<br>mg/L<br>mg/L<br>mg/L   | 1<br>0.001<br>2<br>0.001  | <1<br><0.001<br><2<br>0.015   |

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Methods references and/or additional QA/QC information available on request.

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| Cient:                   |                              | Department of Environm | ent   |                                  | COC Number:                          | 84484               | 9                   |                         |
|--------------------------|------------------------------|------------------------|---|----------------------------------|--------------------------------------|---------------------|---------------------|-------------------------|
| Attention:               |                              | Ms. Leona Hyde         |   |                                  | Date Reported:                       | 2019-0              | 06-28               |                         |
| Client Pro               | ject:                        |                        |   |                                  | Date Submitted:                      | 2019-0              | 06-17               |                         |
| Purchase                 | Order:                       | 2180014303             |   |                                  | Sample Matrix:                       | Water               |                     |                         |
| <u>LAB ID</u><br>1433470 | <u>Supply / D</u><br>WS-S-00 |                        | <u>Client Sample ID</u><br>2019-6401-00-SI-SP | <u>Sample Date</u><br>2019-06-10 | <u>ANALYTE</u><br>Uranium            | <u>UNIT</u><br>mg/L | <u>MRL</u><br>0.001 | <u>RESULT</u><br><0.001 |
|                          | Camp Po                      | nd Brook               |   |                                  | Zinc                                 | mg/L                | 0.01                | <0.01                   |
| Sample comm              | nent:                        |                        |   |                                  | Phosphorus<br>Total Suspended Solids | mg/L<br>mg/L        | 0.002<br>2          | 0.003<br><2             |

Holding time for N-NO2, N-NO3 analysis was exceeded.

Report comment:

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| Cient:  |                            | Department of Enviror | nment              |                | COC Number:                          | 844849      | 9     |        |
|---|----------------------------|-----------------------|--------------------|----------------|--------------------------------------|-------------|-------|--------|
| Attention:  |                            | Ms. Leona Hyde        |                    |                | Date Reported:                       | 2019-0      | 6-28  |        |
| Attention:       Ms. Leona Hyde         Client Project:       Purchase Order:       2180014303        AB ID       Supply / Description       Client Sample ID       S |                            | Date Submitted:       | 2019-0             | 6-17           |                                      |             |       |        |
|   | 2180014303                 |                       |                    | Sample Matrix: | Water                                |             |       |        |
| AB ID   | Supply / D                 | Description           | Client Sample ID   | Sample Date    | ANALYTE                              | <u>UNIT</u> | MRL   | RESULT |
| 433472  |                            |                       | 2019-6403-00-SI-SP | 2019-06-10     | Alkalinity as CaCO3                  | mg/L        | 5     | <5     |
|   | Reid Brook below Tributary |                       |                    |                | Bromide                              | mg/L        | 0.25  | <0.25  |
|   | comment:                   |                       |                    |                | Chloride                             | mg/L        | 1     | 1      |
|   |                            |                       |                    |                | Colour                               | TCU         | 2     | 43     |
|   |                            |                       |                    |                | Conductivity                         | uS/cm       | 5     | 22     |
| -   | D3 analysis was exceeded.  |                       |                    |                | Dissolved Organic Carbon             | mg/L        | 0.5   | 5.2    |
| eport comme   | <u>nt:</u>                 |                       |                    |                | Fluoride                             | mg/L        | 0.10  | <0.10  |
|   |                            |                       |                    |                | Hardness as CaCO3                    | mg/L        | 1     | 5      |
|   |                            |                       |                    |                | N-NH3 (Ammonia)                      | mg/L        | 0.010 | <0.010 |
|   |                            |                       |                    |                | N-NO2 (Nitrite)                      | mg/L        | 0.10  | <0.10  |
|   |                            |                       |                    |                | N-NO3 (Nitrate)                      | mg/L        | 0.10  | <0.10  |
|   |                            |                       |                    |                | рН                                   |             | 1.00  | 6.60   |
|   |                            |                       |                    |                | Sulphate                             | mg/L        | 1     | 2      |
|   |                            |                       |                    |                | Total Dissolved Solids (COND - CALC) | mg/L        | 1     | 14     |
|   |                            |                       |                    |                | Total Kjeldahl Nitrogen              | mg/L        | 0.15  | 0.16   |
|   |                            |                       |                    |                | Total Organic Carbon                 | mg/L        | 0.5   | 5.2    |
|   |                            |                       |                    |                | Turbidity                            | NTU         | 0.1   | 1.8    |
|   |                            |                       |                    |                | Aluminum                             | mg/L        | 0.01  | 0.12   |

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Methods references and/or additional QA/QC information available on request.

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Lab Report Number: 1909867

|                 | Department of Enviro                                     | nment   |                           |  | COC Number:     | 84484  | 9   |   |  |
|-----------------|--|---|---------------------------|--|-----------------|--|---|---|--|
| Attention:      | Ms. Leona Hyde   |   |                           |  | Date Reported:  | 2019-06-28   |   |   |  |
| Client Project: |  |   |                           |  | Date Submitted: |  | 2019-06-17  |   |  |
| Purchase Order: | 2180014303   |   |                           |  | Sample Matrix:  | Water  |   |   |  |
| Sample comment: | 00<br>ok below Tributary<br>y) analysis was exceeded for | Client Sample ID<br>2019-6403-00-SI-SP<br>the entire report. Holding time for N-N | Sample Date<br>2019-06-10 | ANALYTE<br>Antimony<br>Arsenic<br>Barium<br>Boron<br>Calcium<br>Cadmium<br>Cadmium<br>Chromium<br>Chromium<br>Copper<br>Iron<br>Lead<br>Magnesium<br>Manganese<br>Mercury<br>Nickel<br>Potassium<br>Selenium<br>Sodium |                 | UNIT<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L<br>mg/L | MRL<br>0.0005<br>0.001<br>0.01<br>1<br>0.001<br>0.001<br>0.001<br>0.001<br>1<br>0.001<br>0.005<br>1<br>0.001<br>2 | RESULT         <0.0005         <0.01         <0.01         2         <0.001         <0.001         0.001         0.001         <0.001         <0.001         <0.001         <0.001         <0.001         <1         <0.001         <0.006         <1         <0.001         <2 |  |

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Methods references and/or additional QA/QC information available on request.

APPROVAL: \_\_\_\_\_\_\_\_\_Addrine Thomas

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| Cient:                   |         | Department of Environ                           | iment   |                                  | coc                                  | Number:     | 844849                      | 9                           |                                  |
|--------------------------|---------|---|---|----------------------------------|--------------------------------------|-------------|-----------------------------|-----------------------------|----------------------------------|
| Attention:               |         | Ms. Leona Hyde                                  |   |                                  | Date                                 | Reported:   | 2019-0                      | 6-28                        |                                  |
| Client Pro               | ject:   |   |   |                                  | Date                                 | Submitted:  | 2019-0                      | 6-17                        |                                  |
| Purchase                 | Order:  | 2180014303                                      |   |                                  | Sam                                  | ple Matrix: | Water                       |                             |                                  |
| <u>LAB ID</u><br>1433472 | WS-S-00 | <u>Description</u><br>100<br>ok below Tributary | <u>Client Sample ID</u><br>2019-6403-00-SI-SP | <u>Sample Date</u><br>2019-06-10 | <u>ANALYTE</u><br>Uranium<br>Zinc    |             | <u>UNIT</u><br>mg/L<br>mg/L | <u>MRL</u><br>0.001<br>0.01 | <u>RESULT</u><br><0.001<br><0.01 |
| Sample comn              | nent:   |   |   |                                  | Phosphorus<br>Total Suspended Solids |             | mg/L<br>mg/L                | 0.002<br>2                  | 0.005<br><2                      |

Holding time for (Turbidity) analysis was exceeded for the entire report. Holding time for N-NO2, N-NO3 analysis was exceeded.

Report comment:

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Lab Report Number:

1909867

| ient:                | Department of Environm                             | nent               |             | COC Number:                          | 844849                   |       |        |
|----------------------|--|--------------------|-------------|--------------------------------------|--------------------------|-------|--------|
| ttention:            | Ms. Leona Hyde                                     |                    |             | Date Reported:                       | 2019-06-28<br>2019-06-17 |       |        |
| lient Project:       |  |                    |             | Date Submitted:                      |                          |       |        |
| urchase Order:       | 2180014303   |                    |             | Sample Matrix:                       | Water                    |       |        |
| AB ID Supply         | / Description                                      | Client Sample ID   | Sample Date | ANALYTE                              | <u>UNIT</u>              | MRL   | RESULT |
| 433471 WS-S-         |  | 2019-6402-00-SI-SP | 2019-06-10  | Alkalinity as CaCO3                  | mg/L                     | 5     | <5     |
| Tributa              | rry to Reid Brook                                  |                    |             | Bromide                              | mg/L                     | 0.25  | <0.25  |
|                      |  |                    |             | Chloride                             | mg/L                     | 1     | 1      |
| ample comment:       |  |                    |             | Colour                               | TCU                      | 2     | 45     |
| olding time for N-NO | lding time for N-NO2, N-NO3 analysis was exceeded. |                    |             | Conductivity                         | uS/cm                    | 5     | 22     |
|                      |  |                    |             | Dissolved Organic Carbon             | mg/L                     | 0.5   | 5.5    |
| eport comment:       |  |                    |             | Fluoride                             | mg/L                     | 0.10  | <0.10  |
|                      |  |                    |             | Hardness as CaCO3                    | mg/L                     | 1     | 5      |
|                      |  |                    |             | N-NH3 (Ammonia)                      | mg/L                     | 0.010 | <0.010 |
|                      |  |                    |             | N-NO2 (Nitrite)                      | mg/L                     | 0.10  | <0.10  |
|                      |  |                    |             | N-NO3 (Nitrate)                      | mg/L                     | 0.10  | <0.10  |
|                      |  |                    |             | pH                                   |                          | 1.00  | 6.60   |
|                      |  |                    |             | Sulphate                             | mg/L                     | 1     | 2      |
|                      |  |                    |             | Total Dissolved Solids (COND - CALC) | mg/L                     | 1     | 14     |
|                      |  |                    |             | Total Kjeldahl Nitrogen              | mg/L                     | 0.15  | <0.15  |
|                      |  |                    |             | Total Organic Carbon                 | mg/L                     | 0.5   | 5.5    |
|                      |  |                    |             | Turbidity                            | NTU                      | 0.1   | 1.1    |
|                      |  |                    |             | Aluminum                             | mg/L                     | 0.01  | 0.12   |

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Lab Report Number: 1909867

| Cient:   |   | Department of Environ | ment                                   |   |  | COC Number:                  | 84484                               | 9                                     |                                      |
|--|---|-----------------------|--|---|--|------------------------------|-------------------------------------|---------------------------------------|--------------------------------------|
| Attention:   |   | Ms. Leona Hyde        |  |   |  | Date Reported:               | 2019-0                              | )6-28                                 |                                      |
| Attention:<br>Client Project:<br>Purchase Order:<br><u>AB ID Supply / De</u><br>1433471 WS-S-000<br>Tributary t<br>Sample comment: |   |                       |  |   | Date Submitted:                          | 2019-06-17                   |                                     |                                       |                                      |
| Purchase C   | Order:  | 2180014303            |  |   |  | Sample Matrix:               | Water                               |                                       |                                      |
| <u>_AB ID</u><br>1433471   | WS-S-00<br>Tributary                                |                       | Client Sample ID<br>2019-6402-00-SI-SP | <u>Sample Date</u><br>2019-06-10        | ANALYTE<br>Antimony<br>Arsenic<br>Barium |                              | <u>UNIT</u><br>mg/L<br>mg/L<br>mg/L | <u>MRL</u><br>0.0005<br>0.001<br>0.01 | RESULT<br><0.0005<br><0.001<br><0.01 |
| Holding time   | olding time for N-NO2, N-NO3 analysis was exceeded. |                       |  | Boron<br>Calcium<br>Cadmium<br>Chromium |  | mg/L<br>mg/L<br>mg/L<br>mg/l | 0.01<br>1<br>0.0001<br>0.001        | <0.01<br>2<br><0.0001<br><0.001       |                                      |
|  | <u></u>   |                       |  |   | Copper<br>Iron<br>Lead                   |                              | mg/L<br>mg/L<br>mg/L<br>mg/L        | 0.001<br>0.03<br>0.001                | <0.001<br>0.001<br>0.31<br><0.001    |
|  |   |                       |  |   | Magnesium<br>Manganese<br>Mercury        |                              | mg/L<br>mg/L<br>mg/L                | 1<br>0.01<br>0.0001                   | <1<br><0.01<br><0.0001               |
|  |   |                       |  |   | Nickel<br>Potassium<br>Selenium          |                              | mg/L<br>mg/L<br>mg/L                | 0.005<br>1<br>0.001                   | 0.006<br><1<br><0.001                |
|  |   |                       |  |   | Sodium<br>Strontium                      |                              | mg/L<br>mg/L                        | 2<br>0.001                            | <2<br>0.011                          |

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| Cient:   |   | Department of Environment |  |                           |                                      | mber:    | 844849                      |                             |                                  |
|--|---|---------------------------|--|---------------------------|--------------------------------------|----------|-----------------------------|-----------------------------|----------------------------------|
| Attention:                                     |   | Ms. Leona Hyde            |  |                           |                                      | ported:  | 2019-06-28                  |                             |                                  |
| Client Pro                                     | ject:                                     |                           |  |                           | Date Sul                             | bmitted: | 2019-06-17                  |                             |                                  |
| Purchase Order:                                |   | 2180014303                |  |                           |                                      | Matrix:  | Water                       |                             |                                  |
| <u>LAB ID</u><br>1433471                       | <u>Supply / D</u><br>WS-S-00<br>Tributary | 00                        | Client Sample ID<br>2019-6402-00-SI-SP | Sample Date<br>2019-06-10 | <u>ANALYTE</u><br>Uranium<br>Zinc    |          | <u>UNIT</u><br>mg/L<br>mg/L | <u>MRL</u><br>0.001<br>0.01 | <u>RESULT</u><br><0.001<br><0.01 |
| Tributary to Reid Brook <u>Sample comment:</u> |   |                           |  |                           | Phosphorus<br>Total Suspended Solids |          | mg/L<br>mg/L                | 0.002<br>2                  | 0.005<br>2                       |

Holding time for N-NO2, N-NO3 analysis was exceeded.

Report comment:

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