

# Real-Time Water Quality Deployment Report

# **Rattling Brook Network**

January 17, 2019 to March 7, 2019



Government of Newfoundland & Labrador Department of Municipal Affairs and Environment Water Resources Management Division St. John's, NL, A1B 4J6 Canada



## General

- Department of Municipal Affairs and Environment staff monitor the real-time web pages consistently.
- The deployment interval for Big Pond station used in this report is only to maintain consistency compared to Bridge and Plant Discharge stations downstream. Equipment at Big Pond station was actually deployed on December 6, 2018 and remained deployed under the ice for the duration of the winter. The deployment interval for Big Pond will end once ice cover at Big Pond allows for removal.
- A turbidity sensor failure at Rattling Brook below Bridge required a mid-deployment switch of equipment on January 30, 2019.
- Hydrometric data included in this report is provisional and used only for illustrative purposes. Corrected and finalized data may be retrieved from the Water Survey of Canada website (http://www.ec.gc.ca/rhc-wsc/)\*.

## Maintenance and Calibration of Instrument

- As part of the Quality Assurance and Quality Control protocol (QAQC), an assessment of the reliability of data recorded by an instrument is made at the beginning and end of the deployment period. The procedure is based on the approach used by the United States Geological Survey.
  - Upon deployment, a QA/QC Sonde is temporarily deployed *in situ*, adjacent to the Field Sonde. Depending on the degree of difference between each parameter from the Field and QAQC sondes a qualitative rank is assigned (See Table 1). The possible ranks, from most to least desirable, are: Excellent, Good, Fair, Marginal, and Poor. A grab sample is also taken for additional confirmation of conditions at deployment and to allow for future modelling studies.
  - At the end of a deployment period, a freshly cleaned and calibrated QAQC Sonde is placed *in situ*, adjacent to the Field Sonde. Values are compared between all parameters and differences are ranked for placement in Table 1.

Station	Date	Action	Comparison Ranking				
			Temperature	pН	Conductivity	<b>Dissolved Oxygen</b>	Turbidity
Rattling Brook Big Pond	December 6, 2018	Deployment	Excellent	Excellent	Excellent	NA	Excellent
	-	Ongoing	NA	NA	NA	NA	NA
Rattling Brook below Bridge	January 18, 2019	Deployment	Excellent	Excellent	Excellent	Excellent	Excellent
	March 7, 2019	Removal	Excellent	Good	Good	Excellent	Marginal
Rattling Brook below Plant Discharge	January 18, 2019	Deployment	Excellent	Excellent	Excellent	Good	Marginal
	March 7, 2019	Removal	Excellent	Fair	Good	Fair	Good

Table 1:	Qualitative	QAQC	Ranking
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# **Data Interpretation**

## Temperature

Water Temperature is a major factor used to describe water quality. Temperature has major implications on both the ecology and chemistry of a water body, governing processes such as the metabolic rate of aquatic plants and animals and the degree of dissolved oxygen saturation. Rattling Brook

Temperature g Pond 0 3. Temperature (°C) 2 3 -2 Plant 0 108.5 Big Pond 108.4 108.3 85 Stage (m) 84 83 73.0 72.8 72.6 72.4 Mar 02 -Mar 04-Mar 10-Mar 06 -Jan 27 Feb 02 eb 04 Feb 14 eb 20 eb 22 eb 26 Feb 28 Mar 08 an 21 an 23 eb 18 eb 24 la Feb B B E B Median Station Mean Min Max **Big Pond** 1.16 1.08 0.33 1.91 0.71 0.65 -0.04 2.70 **Below Bridge** 

 Water temperatures at Big Pond station differed substantially from those observed downstream at Bridge and Plant Discharge stations. Trend lines show a general increase in water temperature at Big Pond station from mid-January to March whereas trend lines show a decrease at downstream stations. This difference is likely due to the degree of ice cover providing insulation from cold air temperatures at Big Pond station.

0.41

Below Plant Discharge

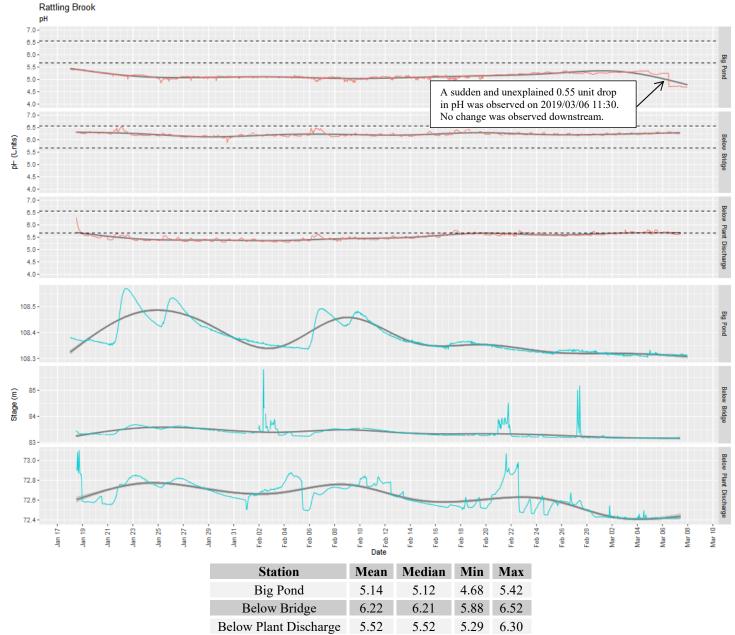
0.06

-0.07

2.78

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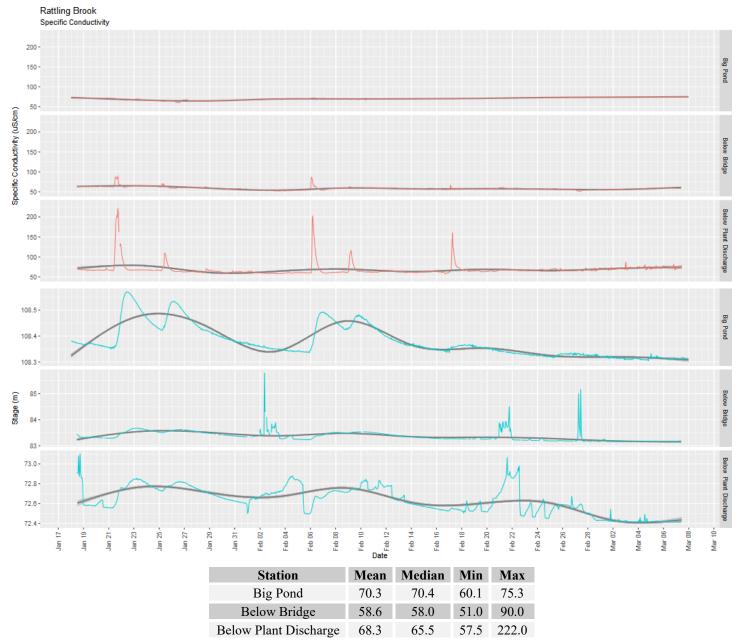
pH is used to give an indication of the acidity or basicity of a solution. A pH of 7 denotes a neutral solution while lower values are acidic and higher values are basic. Technically, the pH of a solution indicates the availability of protons to react with molecules dissolved in water. Such reactions can affect how molecules function chemically and metabolically.



- pH values tend to show stability during the winter season that is not observed during spring, summer, and fall. Key differences are the decrease in biological activity during winter and the reduced volume of freshwater entering the river system as runoff.
- Most pH values fell below site-specific guidelines at Big Pond and Plant Discharge stations whereas most were within site-specific guidelines at Bridge station.

#### **Specific Conductivity**

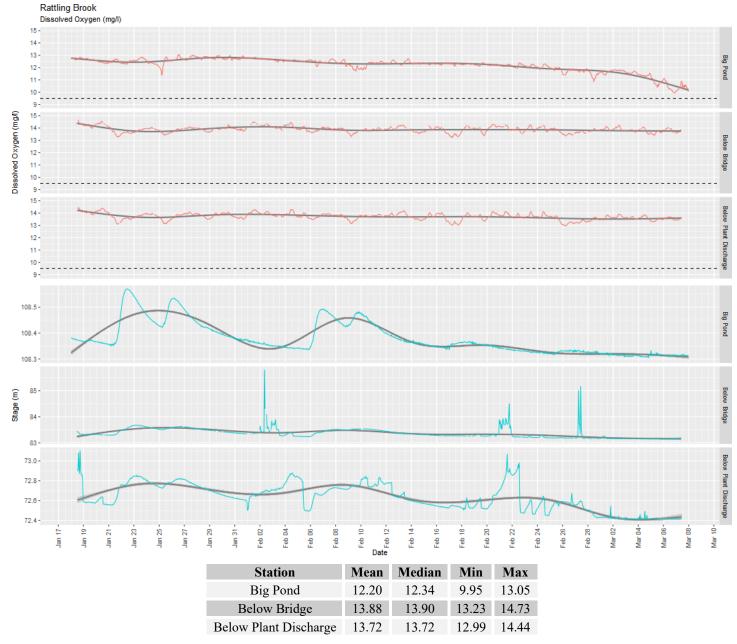
Conductivity relates to the ease of passing an electric charge – or resistance – through a solution. Conductivity is highly influenced by the concentration of dissolved ions in solution: distilled water has zero conductivity (infinite resistance) while salty solutions have high conductivity (low resistance). Specific Conductivity is corrected to 25°C to allow comparison across variable temperatures.



Specific conductivity at Big Pond station was very stable from January to March. Bridge station was found to be slightly lower in conductivity with a presence of conductivity events that were absent at Big Pond station (an effect of ice cover). The largest conductivity events were seen at Plant Discharge station due to the cumulative effect of dissolved solids entering Rattling Brook along the length of the river.

### **Dissolved Oxygen**

Dissolved oxygen is a metabolic requirement of aquatic plants and animals. The concentration of oxygen in water depends on many factors, especially temperature – the saturation of oxygen in water is inversely proportional to water temperature. Oxygen concentrations also tend to be higher in flowing water compared to still, lake environments. Low oxygen concentrations can give an indication of excessive decomposition of organic matter or oxidation reactions.



 Ice cover prevents the exchange of dissolved gasses with the atmosphere. As a result, dissolved oxygen is slowly consumed at Big Pond station through biological processes and inorganic oxidation reactions. At Bridge and Plant Discharge stations, dissolved oxygen is not depleted – gas is exchanged as water passes over falls and riffles.

## Turbidity

Turbidity is typically caused by fine suspended solids such as silt, clay, or organic material. Consistently high levels of turbidity tend to block sunlight penetration into a waterbody, discouraging plant growth. High turbidity can also damage the delicate respiratory organs of aquatic animals and cover spawning areas.

Rattling Brook Turbidity 400 300 Big 200 Pond 100 0 400 Sensor failure and **Turbidity (NTU**) 300 subsequent replacement. Below 200 Bridge 100 0 400 Below Plant Discharge 300 200 100 108.5 Big Pond 108.4 108.3 Arrows indicate instances of stage level increase 85 Stage (m) and turbidity increase. Bridge 84 83 73.0 72.8 Plant Disc 72.6 72.4 Feb 22 -Feb 28 ġ Mar 04ģ ģ Jan 27 5 eb 02 eb 04 Feb 14 eb 24 eb 26 an 21 Feb 12 Ind E Feb 9 8 Mar Mar Mar Mar Mean Median Min Max Station **Big Pond** 0.0 0.0 0.0 56.2 11.2 Below Bridge 13.1 0.0 398.0 Below Plant Discharge 3.6 1.0 0.3 335.8

 Turbidity at Big Pond station was low and showed very little variation during this deployment period due to complete ice cover. The lack of complete ice cover downstream at Bridge and Plant Discharge stations allowed for influence from precipitation and silt deposition into the river. As a result, variability was notable, especially during stage level changes.

## Appendix

