

A photograph of a water treatment plant, showing large circular tanks and metal walkways. The image is overlaid with a grid of semi-transparent grey squares. The text is positioned in the bottom left corner of the slide.

Technical Session on Arsenic in the Environment:

Sources, Guidelines, Risks and Treatment

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Overview of Arsenic

- Widely distributed in the earth's crust
- In water by dissolution of ores/minerals
- Concentrated in areas based on bedrock type
- Becomes available by erosion or by movement within fractures

Sources of Arsenic

- Natural Sources of arsenic
 - Contained in 250 naturally occurring minerals (particularly sulphides)
 - Present in volcanic-derived sediment
 - Present in rivers and streams
 - Aquifer materials high in iron oxides and sulphide minerals

Sources of Arsenic

- Human Sources of arsenic
 - Mineral extraction and processing
 - Glass manufacturing
 - Wood preservation
 - Pesticide production and application
 - Waste burning and leachate
 - Coal/oil production and processing

Forms of Arsenic

- Metallic arsenic – not absorbed by the body
- Organic arsenic – primary form of arsenic in fish and shellfish (shows little to no toxicity)
- Inorganic arsenic – Arsenate (As^{+5}) and Arsenite (As^{3+}) are the most prevalent forms in drinking water

Arsenic in Soil and Bedrock

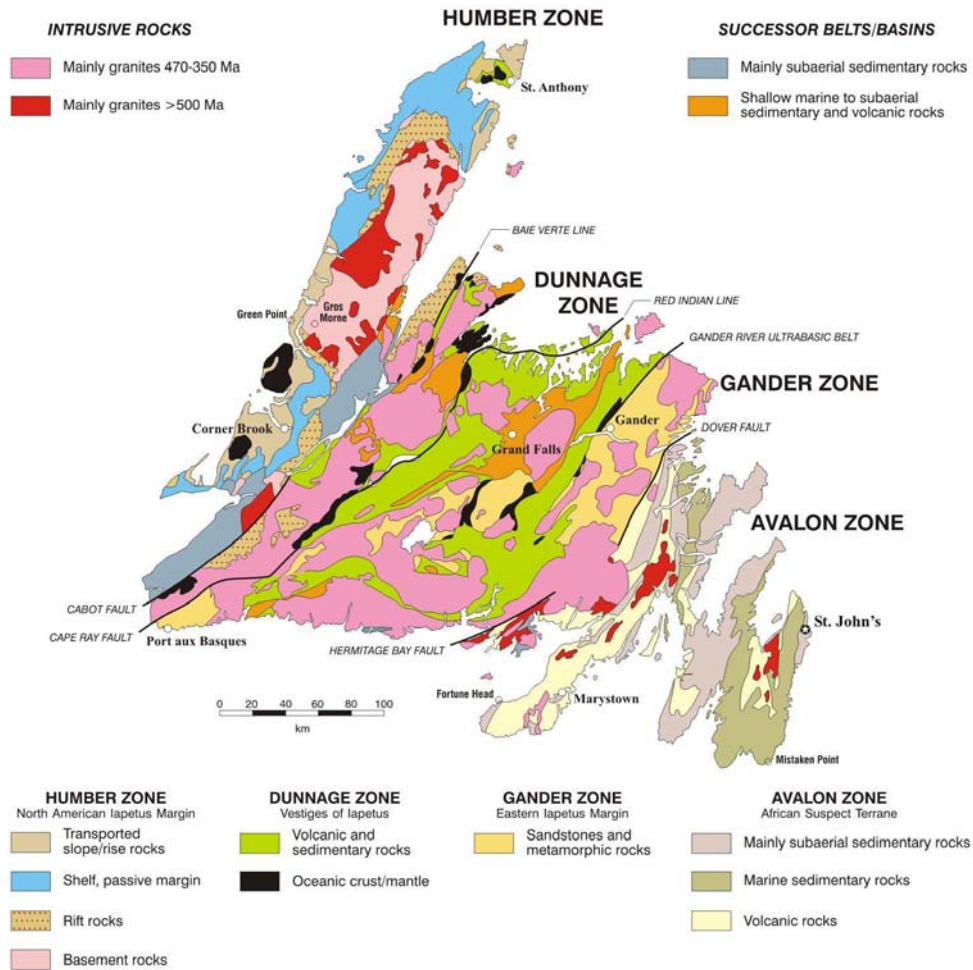
- Soil:
 - Average concentrations in Canada range from 4.8 to 13.6 mg/kg
 - Higher in industrialized areas, mining operations, smelters
 - Higher in areas based on bedrock type
 - Soil quality guideline for any land use type is 12 mg/kg (based on soil ingestion)

- Bedrock:
 - Typical bedrock includes volcanic and sedimentary rocks
 - In NL, this includes Baie Verte area, Lewisporte and Port aux Basques



Geological Survey
Department of
Natural Resources

GENERALIZED INTERPRETIVE MAP- NEWFOUNDLAND APPALACHIANS



Arsenic in Groundwater

- Mobility largely controlled by redox conditions, pH, biological activities, and adsorption/desorption reactions
- Aquifers act as a “vessel” enabling the migration of arsenic
- Tasteless and odourless
- In Canadian drinking water, generally less than 0.005 mg/L, but is dependant on many factors!

Arsenic Mobility Profile

- Arsenates - As(V)
 - Usually occurs in aqueous environments that are oxygenated (DO>1ppm) - i.e., HnAsO_4^{n-3}
 - predominantly in aerobic soils or well-oxygenated surface waters
- Arsenites – As(III)
 - Usually occurs in aqueous environments that are high in sulphides or methane (DO<1ppm) - i.e., HnAsO_3^{n-3}
 - Exists in slightly reduced soils
 - More of a concern with respect to human exposure due to its toxicity and mobility
 - predominantly in reducing conditions (deep lake sediments and groundwater)
 - 4 to 10 times more soluble in water than As(V)

Arsenic in Groundwater

- The occurrence and variability of arsenic in groundwater could be related to the following factors:
 - Distribution of arsenic in soil and rocks that are part of the groundwater flow system
 - Characteristics that influence the solubility and transport of arsenic in groundwater
 - Associations between land use, geology and groundwater flow patterns

Guidelines

- Health Canada drinking water guideline of 0.010 mg/L
- Based on municipal and residential scale treatment achievability
- Guideline is based on lifetime exposure (70 years)
- Takes into consideration the ability to measure and remove arsenic from drinking water supplies

As Levels in Canadian Water

■ PEI

- Groundwater levels ranged from 0.0001 to 0.026 mg/L between 1986 and 2002

■ Alberta

- Groundwater levels ranged from 0.0001 to 1 mg/L in treated groundwater and surface water supplies between 1980 and 2002

■ NL

- Groundwater levels ranged from 0.006 to 0.288 mg/L in public water supplies in 2002
- Groundwater levels ranged from 0.001 to 0.368 mg/L in public schools in 2002.
- 19% schools exceeded guideline of 0.010 mg/L

Arsenic in Groundwater in NL

- NL data strongly based on lake-sediment geochemistry
- Well defined pattern in groundwater chemistry and underlying bedrock compared to wells installed in soil or dug wells
- Related to fracture zones

Areas of Potential Arsenic Concentration in Well Water

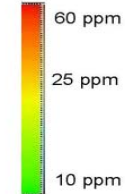
Areas where high arsenic is found in lake sediments are more likely to have arsenic in drilled wells that encounter bedrock.

Legend

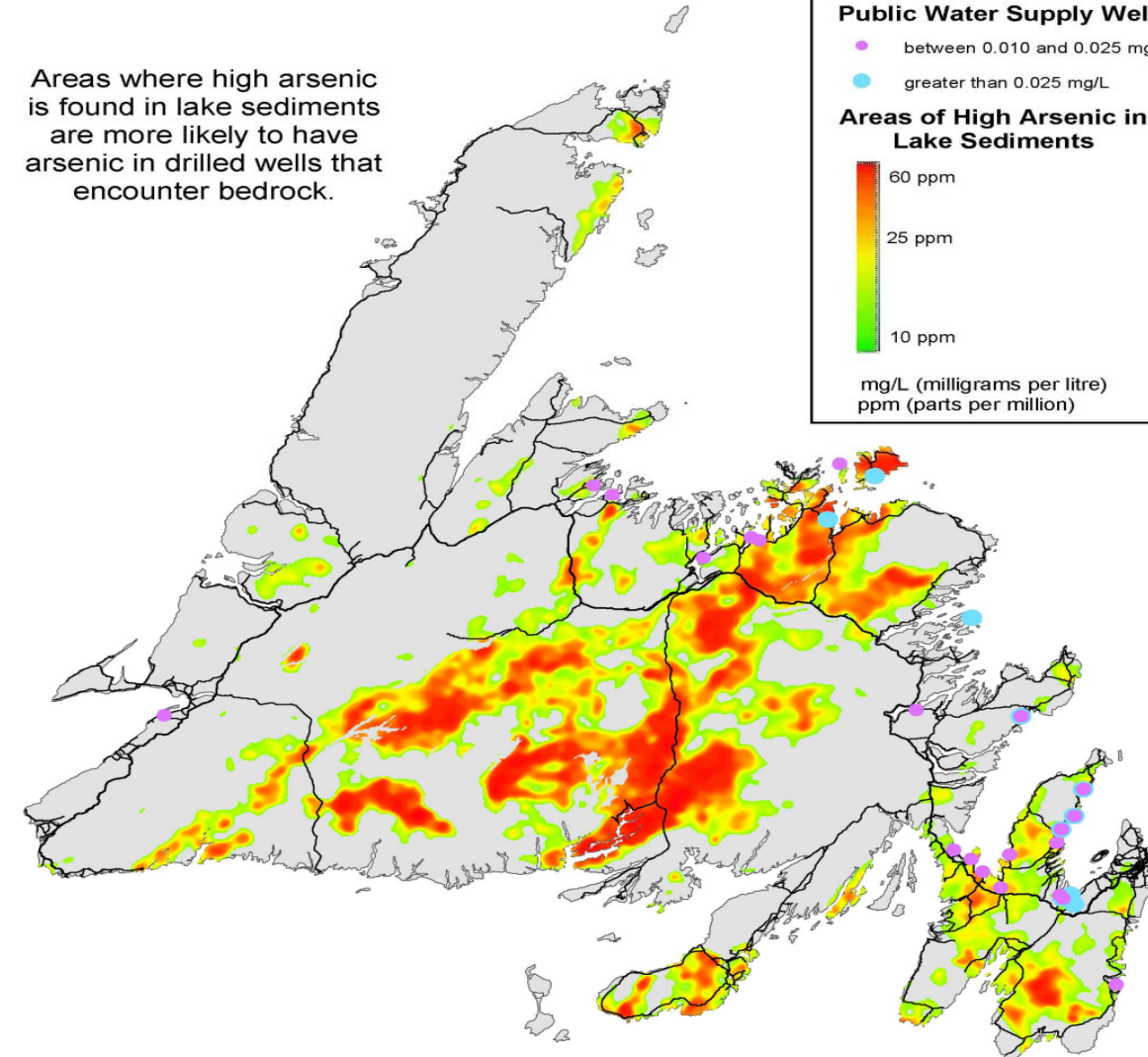
Public Water Supply Wells

- between 0.010 and 0.025 mg/L
- greater than 0.025 mg/L

Areas of High Arsenic in Lake Sediments



mg/L (milligrams per litre)
ppm (parts per million)



Government of Newfoundland & Labrador
Department of Environment
Department of Mines and Energy

0 25 50 100 150 Kilometers

Arsenic in Groundwater in NL

- Approximately 27 % of Newfoundland obtain their drinking water from groundwater
- Public supplies routinely tested by government officials
- Conducting sampling since 2001
- Sampling conducted in NL communities since 2006 ranged from 0.011 mg/L to 0.027 mg/L
- Detected in school water supplies in St. Brendan's and Wings Point
- Detected in public water supplies including the northwest Avalon Peninsula and Fogo Island

Arsenic in Groundwater in NL

- Some communities have been on a “non-consumption order” by the Dept. of Environment
- Some communities have installed treatment systems or has turned to alternate sources
- No arsenic exceedances reported for Labrador
- Approximately 20,000 privately drilled wells have not been tested (as of 2002)
- Approximately 500 newly drilled wells installed each year

Arsenic Exposure

- Arsenic is a human cancer-causing agent
- For most Canadians, primary source is food, followed by drinking water, soil and air
- Absorbed by the body when ingested and distributed by the blood stream
- Exposure by dermal contact or inhalation it not considered to be significant
- Studies do not indicate greater risks to children or pregnant women in comparison to adults

Arsenic Exposure

- Food sources:
 - Adult 42 ug
 - Child 14.9 ug
 - Vegetation < Fish < Shellfish

- Inhalation:
 - Mean ambient level for 11 Canadian cities 0.001 ug/m³
 - Higher concentrations near smelters or industrial activities

- Drinking water:
 - Adult < 7.5 ug (based on an intake of 1.5 L)
 - Child < 3.5 ug (based on an intake of 0.7 L)

Effects due to Exposure

- Long term effects due to As in drinking water include:
 - Skin changes: pigmentation changes and thickening
 - Nausea and diarrhea
 - Decreased production of blood cells
 - Abnormal heart beat
 - Numbness in the hands and feet
- Short term effects include:
 - Abdominal pain, muscular pain
 - Flush skin or rash
 - Vomiting and diarrhea
 - Numbness in hands and feet

Effects due to Long Term Exposure

- Other Long term effects due to As exposure via other sources include:
 - Cancer of the skin, lungs, urinary bladder, kidneys

Treatment Options

- Typical concentration of arsenic in groundwater is 0.005 mg/L
- Treatment required to obtain a level of <0.010 mg/L
- Treatment based upon scale of drinking water supply (residential vs. municipal)
- Short term vs. long term exceedances

Treatment Options

- Treatment type based on characteristics of water supply (pH, iron etc.)
- pH = 4 to 10
 - As (V) has –ve charge and is easier to remove
 - As(III) is neutral
- Need to convert As(III) to As(V) by oxidation
- May need to adjust pH (remove TDS and competing ions), often referred to as a “pre-treatment step”

Treatment Options

- Key Considerations when choosing a treatment option:
 - System maintenance
 - Cost effectiveness
 - Difficulty of use
 - Multi-use system?

- Before any treatment system is considered, must understand the water chemistry (pH, competing ions, organic content).

Municipal Scale

- Coagulation/filtration combined with “pre-treatment”
 - Can reduce As to 0.003 -0.005 mg/L
 - Also removes suspended and dissolved solids

- Lime softening
 - Can reduce As to 0.001-0.003 mg/L
 - Good for addressing hard water
 - More expensive

- Activated alumina adsorption with microfiltration
 - Can reduce As to <0.01mg/L (95% effective)
 - Involves chemical handling

Municipal Scale

- Treatment types discussed are best suited for larger scale systems
- Can create backwash, contaminated filters for disposal etc.
- Other options include:
 - Reverse osmosis
 - can remove up to 85%
 - requires a lot of water
 - results in a lot of wastewater and arsenic rich brine
 - Not recommended where water resources are scarce
 - Manganese greensand filtration (not a high removal rate, requires iron)

Residential Scale

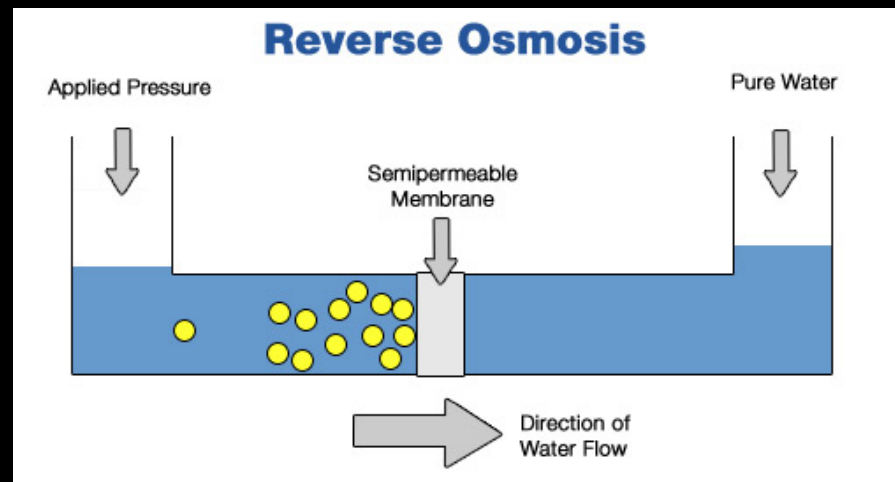
- Applies to private potable wells

- Have well tested
 - Collect sample at point of entry or point of source
 - Prior to any treatment system or filters

- Arsenic is not removed by boiling the water!

Residential Scale

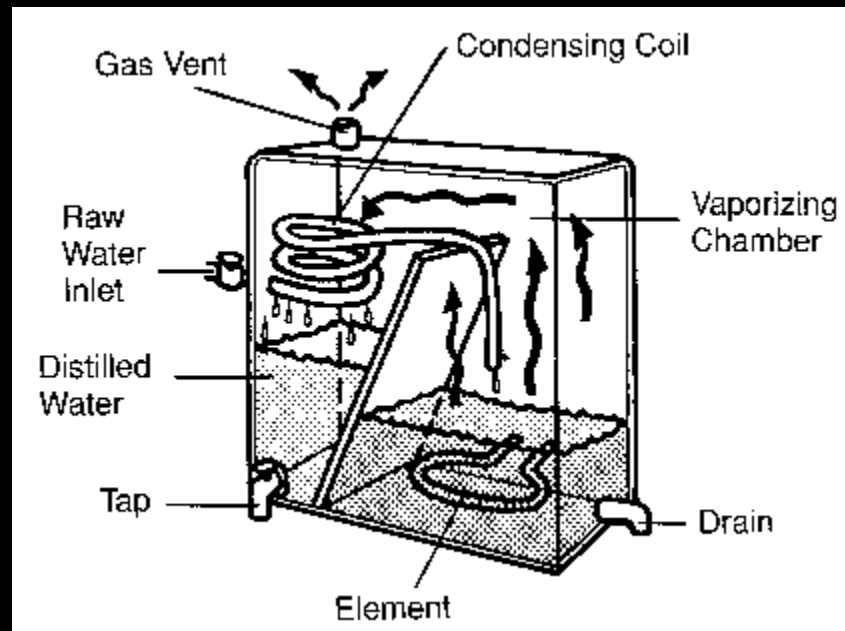
- Reverse osmosis
 - Requires large quantity of influent water
 - 5 gallons of treated water = 40 to 90 gallons of wastewater
 - Removes other dissolved minerals (98%)
 - Can be installed at point of use or point of entry



Residential Scale

■ Steam distillation

- Heat also kills bacteria that may exist, as well as all other minerals
- Installed at point of use, but can be noisy!
- Low maintenance



Residential Scale

- Adsorption
 - Adsorbent filter (carbon or charcoal)
 - in place within the water line prior to point of use
 - Remove other metals, but not bacteria

Prevention and Control

- Inquire about your potable water supply
- Implementation of a monitoring plan in the interim
- Installation of a Treatment System

- Alternate water sources:
 - Connect to public distribution system
 - Look for new Groundwater source

- Education:
 - Sample your residential well
 - Communication with general public

Questions?????
