
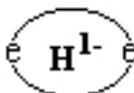

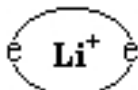
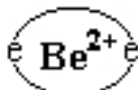

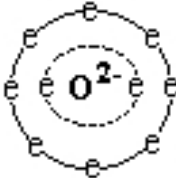
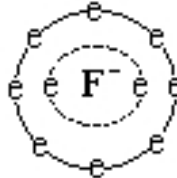










## ANSWERS

 H					 H	 He	
 Li	 Be	Does Not Form an Ion B	Does Not Form an Ion C	 N	 O	 F	 Ne
 Na	 Mg	 Al	Does Not Form an Ion Si	 P	 S	 Cl	 Ar

Note that hydrogen forms both a positive and a negative ion (very unique). Also, the noble gases were included in this chart so that the ions that have formed can be compared its nearest noble gas.