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Mines

# RECONNAISSANCE-LEVEL LITHOGEOCHEMICAL SAMPLING OF IGNEOUS ROCKS FROM THE AVALON TERRANE, EASTERN NEWFOUNDLAND (NTS MAP AREAS 1M/3, 10, 1N/5, 6, 10, 11 AND 12)

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#### SUMMARY

This open file data release consists of whole-rock geochemical data from 35 rock samples collected from various parts of the Avalon Terrane, eastern Newfoundland (Figure 1, NTS map areas 1M/3, 10, 1N/5, 6, 10, 11 and 12). The bedrock geology of the core of the Avalon Peninsula was compiled by McCartney (1967) and the entire Avalon Peninsula (and northwest to include Random Island) was compiled by King (1988). The bedrock geology of the Terrenceville map (1M/10) was completed by O'Brien *et al.* (1984), and the Marystown map was completed by Strong *et al.* (1977). All samples included in this report were collected by A.J. Mills (2022; four samples in 2021). This open file data release provides no interpretation.

## NOTES ON THE DATABASE

All data are reported in Zone 22 UTM coordinates. Samples are prefixed by the year and initials of the geologist who collected them. Appendix A contains brief sample descriptions, location data, as well as major- and trace-element whole-rock geochemical analytical data for the samples collected. Analytical duplicates were selected at random and inserted at a frequency of one in 20 (Appendix B). In addition, a number of reference materials (Standards) were analyzed for quality assurance (Appendix C). Details of the analytical methods used are provided by Finch *et al.* (2018) and summarized in Table 1. The data are available in digital format (*i.e.*, comma separated value files; \*.csv) through the links provided in the Appendices section.

Major-element compositions (plus Ba, Be, Cr, Sc and Zr) were analyzed using ICP-OES methods, following lithium tetraborate and metaborate fusion. REE and selected trace elements were determined by ICP-MS analysis after lithium tetraborate and metaborate fusion. Other trace elements (As, Cd, Co, Cu, Li, Mo, Ni, Pb, Rb, S, V and Zn) were analyzed by ICP-OES after total



Figure 1. Location of study area.

4-acid digestion (note that Rb in the Burin Peninsula samples was analyzed by ICP-MS only). Volatiles are represented as loss-onignition (LOI) at 1000°C, which represents the breakdown of all minerals and release of all volatiles. The ferrous-iron content (FeO) of silicate rocks is determined by the Wilson Method (Wilson, 1960), as outlined by Finch et al. (2018). For silver analysis, 0.5 g of sample powder was weighed into a 15 ml digestion tube with 2 ml of concentrated nitric acid, and digested for two hours and then analyzed by ICP-OES (Finch et al., 2018). Flouride content is determined as described by Ficklin (1970) and Finch et al. (2018).

Major elements are reported in weight percent (wt. %), and minor and trace elements are reported in parts per million

Abbreviation	Explanation
-99	Sample was not analyzed for that element
Avg	Average value
Dup	Duplicate analysis
Fe <sub>2</sub> O <sub>3</sub> T	Total measured iron
ICP-OES-4-ACID	Inductively Coupled Plasma Optical Emission Spectrometry following HF-HCl-HNO <sub>3</sub> -HClO <sub>4</sub> acid digestion
ICP-OES-FUS	Inductively Coupled Plasma Optical Emission Spectrometry following lithium metaborate/tetraborate fusion
ICP-OES-HNO <sub>3</sub>	Inductively Coupled Plasma Optical Emission Spectrometry following nitric acid digestion
ICP-MS-FUS	Inductively Coupled Plasma Mass Spectrometry following lithium metaborate/tetraborate fusion
ISE	Ion-selective electrode
LCL	Lower control limit
LOI	Loss-on-ignition
negative detection limit	Below detection limit
pct	Percent
ppm	Parts per million
Rec_Val	Recommended value
UCL	Upper control limit
wt_pct	Weight percent

#### Table 1. List of abbreviations

(ppm). A negative number indicates that the concentration was below the detection limit (e.g., - 0.01 indicates the measured value was below the detection limit of 0.01). Detection limits are listed for each element in the .csv files.

Mg# was determined by the formula: Mg# =  $(MgO/40.312)/((MgO/40.312)+(FeO^{T}/71.847))*100.$ 

Ishikawa Alteration Index (see Large et al., 2001):  $AI = 100*(K_2O + MgO)/(K_2O + MgO + Na_2O + CaO)$ 

Chemical Index of Alteration (Nesbitt and Young, 1982):  $CIA = [Al_2O_3/(Al_2O_3 + CaO^* + Na_2O + K_2O)] *100$ 

Within the Duplicates Table (Appendix B): %\_difference = [(OriginalValue - Lab Split Value)/Original Value] \* 100.

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Element	Analytical Method	Preparation/Digestion
SiO <sub>2</sub> , Al <sub>2</sub> O <sub>3</sub> , Fe <sub>2</sub> O <sub>3</sub> T, MgO, CaO, Na <sub>2</sub> O, K <sub>2</sub> O, TiO <sub>2</sub> , MnO, P <sub>2</sub> O <sub>5</sub> , Ba, Be, Cr, Sc, Zr	Inductively Coupled Plasma Optical Emission Spectrometry (ICP-OES)	50-50 Lithium Tetraborate Lithium Metaborate Fustion
FeO	Titration	NH <sub>4</sub> VO <sub>3</sub> , HF, H <sub>2</sub> SO <sub>4</sub> , HSPO <sub>4</sub>
As, Cd, Co, Cu, Li, Mo, Ni, Pb, Rb, S, V, Zn	Inductively Coupled Plasma Optical Emission Spectrometry (ICP-OES)	Hf-HCl-HNO <sub>3</sub> -HClO <sub>4</sub> (total digestion)
Bi, Ce, Cs, Dy, Er, Eu, Ga, Gd, Ge, Hf, Ho, La, Lu, Nb, Nd, Pr, Rb, Sm, Sn, Sr, Ta, Tb, Th, Ti, Tm, U, W, Y, Yb	Inductively Coupled Plasma Mass Spectrometry (ICP-MS)	50-50 Lithium Tetraborate Lithium Metaborate Fusion
F	Ion Selective Electrode (ISE)	Na <sub>2</sub> CO <sub>3</sub> and KNO <sub>3</sub> fusion
Ag	Inductively Coupled Plasma Optical Emission Spectrometry ICP-OES)	HNO <sub>3</sub> digestion
LOI	Gravimetric (Grav) at 1000°C	None

 Table 2. Analytical methods for geochemical analyss

entist Javier Alvaro for sampling on the Avalon Peninsula, and by colleague James Conliffe and assistant Noah Slaney on the Burin Peninsula. Thanks also go to Pauline Honarvar for a thorough review of the data for quality control.

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#### APPENDICES

Appendices A–C are available as digital comma-separated files (.csv) through this link.

**APPENDIX A:** Major-element and Trace-element Data

**APPENDIX B:** Major-element and Trace-element Data for Duplicates

**APPENDIX C:** Major-element and Trace-element Data for Standards, with Certified Reference Materials